

# Ethyl

## 20-hydroxy-3,6,9,12,15,18-hexaoxaicosan-1-oate

Inchi: InChI=1S/C16H32O9/c1-2-25-16(18)15-24-14-13-23-12-11-22-10-9-21-8-7-20-6-5-19-4-3

InchiKey: ZNXSBGRKYILDKF-UHFFFAOYSA-N

Formula: C16H32O9

SMILES: CCOC(=O)COCCOCCOCCOCCOCCOCCO

Mol. weight [g/mol]: 368.42

## Physical Properties

Property code	Value	Unit	Source
gf	-916.90	kJ/mol	Joback Method
hf	-1563.92	kJ/mol	Joback Method
hfus	51.20	kJ/mol	Joback Method
hvap	91.51	kJ/mol	Joback Method
log10ws	0.83		Crippen Method
logp	-0.359		Crippen Method
mvol	284.830	ml/mol	McGowan Method
pc	1353.63	kPa	Joback Method
rinpol	2649.90		NIST Webbook
rinpol	2649.90		NIST Webbook
tb	868.47	K	Joback Method
tc	1064.12	K	Joback Method
tf	536.44	K	Joback Method
vc	1.083	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	929.17	J/molxK	868.47	Joback Method
cpg	944.35	J/molxK	901.08	Joback Method
cpg	958.17	J/molxK	933.69	Joback Method
cpg	970.62	J/molxK	966.30	Joback Method
cpg	981.63	J/molxK	998.91	Joback Method
cpg	991.17	J/molxK	1031.52	Joback Method
cpg	999.20	J/molxK	1064.12	Joback Method
dvisc	0.0000950	Paxs	536.44	Joback Method

dvisc	0.0000423	Paxs	591.78	Joback Method
dvisc	0.0000217	Paxs	647.12	Joback Method
dvisc	0.0000123	Paxs	702.45	Joback Method
dvisc	0.0000076	Paxs	757.79	Joback Method
dvisc	0.0000050	Paxs	813.13	Joback Method
dvisc	0.0000035	Paxs	868.47	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R578084&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R578084&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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