

Cyclohexane, 1,1'-(2-methyl-1,3-propanediyl)bis-

Other names:	1,3-Dicyclohexyl-2-methylpropane Propane, 1,3-dicyclohexyl-2-methyl-
Inchi:	InChI=1S/C16H30/c1-14(12-15-8-4-2-5-9-15)13-16-10-6-3-7-11-16/h14-16H,2-13H2,1H3
InchiKey:	MUNDOJMYSNYILP-UHFFFAOYSA-N
Formula:	C16H30
SMILES:	CC(CC1CCCCC1)CC1CCCCC1
Mol. weight [g/mol]:	222.41
CAS:	2883-08-1

Physical Properties

Property code	Value	Unit	Source
gf	130.30	kJ/mol	Joback Method
hf	-270.21	kJ/mol	Joback Method
hfus	17.34	kJ/mol	Joback Method
hvap	51.68	kJ/mol	Joback Method
log10ws	-5.58		Crippen Method
logp	5.563		Crippen Method
mcvol	214.580	ml/mol	McGowan Method
pc	1832.54	kPa	Joback Method
tb	604.14	K	Joback Method
tc	824.40	K	Joback Method
tf	269.84	K	Joback Method
vc	0.791	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	601.49	J/mol×K	604.14	Joback Method
cpg	628.82	J/mol×K	640.85	Joback Method
cpg	654.44	J/mol×K	677.56	Joback Method
cpg	678.42	J/mol×K	714.27	Joback Method
cpg	700.82	J/mol×K	750.98	Joback Method
cpg	721.69	J/mol×K	787.69	Joback Method
cpg	741.10	J/mol×K	824.40	Joback Method

dvisc	0.0121478	Pa×s	269.84	Joback Method
dvisc	0.0031999	Pa×s	325.56	Joback Method
dvisc	0.0012448	Pa×s	381.27	Joback Method
dvisc	0.0006161	Pa×s	436.99	Joback Method
dvisc	0.0003575	Pa×s	492.71	Joback Method
dvisc	0.0002317	Pa×s	548.42	Joback Method
dvisc	0.0001627	Pa×s	604.14	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.39522e+01
Coeff. B	-4.37988e+03
Coeff. C	-9.91510e+01
Temperature range (K), min.	419.68
Temperature range (K), max.	606.04

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C2883081&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀ws:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
p_c:	Critical Pressure
p_{vap}:	Vapor pressure
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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