

# Propionic acid, 3-iodo-, tridecyl ester

<b>Inchi:</b>	InChI=1S/C16H31IO2/c1-2-3-4-5-6-7-8-9-10-11-12-15-19-16(18)13-14-17/h2-15H2,1H3
<b>InchiKey:</b>	ADHQYZDCEIWAQ-UHFFFAOYSA-N
<b>Formula:</b>	C16H31IO2
<b>SMILES:</b>	CCCCCCCCCCCCOC(=O)CCI
<b>Mol. weight [g/mol]:</b>	382.32

## Physical Properties

Property code	Value	Unit	Source
gf	-91.96	kJ/mol	Joback Method
hf	-541.50	kJ/mol	Joback Method
hfus	44.39	kJ/mol	Joback Method
hvap	69.74	kJ/mol	Joback Method
log10ws	-6.33		Crippen Method
logp	5.666		Crippen Method
mvol	269.560	ml/mol	McGowan Method
pc	1341.76	kPa	Joback Method
rinpol	2258.00		NIST Webbook
tb	734.91	K	Joback Method
tc	922.58	K	Joback Method
tf	400.30	K	Joback Method
vc	1.044	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	744.87	J/molxK	734.91	Joback Method
cpg	761.63	J/molxK	766.19	Joback Method
cpg	777.52	J/molxK	797.47	Joback Method
cpg	792.58	J/molxK	828.74	Joback Method
cpg	806.83	J/molxK	860.02	Joback Method
cpg	820.30	J/molxK	891.30	Joback Method
cpg	833.02	J/molxK	922.58	Joback Method
dvisc	0.0017058	Paxs	400.30	Joback Method
dvisc	0.0007850	Paxs	456.07	Joback Method

dvisc	0.0004278	Paxs	511.84	Joback Method
dvisc	0.0002627	Paxs	567.61	Joback Method
dvisc	0.0001760	Paxs	623.37	Joback Method
dvisc	0.0001259	Paxs	679.14	Joback Method
dvisc	0.0000948	Paxs	734.91	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406244&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406244&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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