

# Benzene, 1,3-bis(1,1-dimethylethyl)-2-methoxy-5-methyl-

Other names:	2,6-bis-(tertibutyl)-4-methyl anisole
Inchi:	InChI=1S/C16H26O/c1-11-9-12(15(2,3)4)14(17-8)13(10-11)16(5,6)7/h9-10H,1-8H3
InchiKey:	TVWAVADIILZVIY-UHFFFAOYSA-N
Formula:	C16H26O
SMILES:	COc1c(C(C)(C)C)cc(C)cc1C(C)(C)C
Mol. weight [g/mol]:	234.38
CAS:	1518-53-2

## Physical Properties

Property code	Value	Unit	Source
gf	68.04	kJ/mol	Joback Method
hf	-321.17	kJ/mol	Joback Method
hfus	16.43	kJ/mol	Joback Method
hvap	55.29	kJ/mol	Joback Method
log10ws	-4.69		Crippen Method
logp	4.599		Crippen Method
mcvol	218.410	ml/mol	McGowan Method
pc	1670.06	kPa	Joback Method
rinpol	1496.00		NIST Webbook
rinpol	1496.00		NIST Webbook
tb	623.06	K	Joback Method
tc	835.09	K	Joback Method
tf	361.13	K	Joback Method
vc	0.820	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	580.40	J/molxK	623.06	Joback Method
cpg	667.94	J/molxK	799.75	Joback Method
cpg	652.58	J/molxK	764.41	Joback Method
cpg	636.21	J/molxK	729.07	Joback Method
cpg	618.76	J/molxK	693.74	Joback Method
cpg	600.17	J/molxK	658.40	Joback Method

cpg	682.34	J/molxK	835.09	Joback Method
dvisc	0.0000884	Paxs	623.06	Joback Method
dvisc	0.0001167	Paxs	579.40	Joback Method
dvisc	0.0001613	Paxs	535.75	Joback Method
dvisc	0.0002360	Paxs	492.09	Joback Method
dvisc	0.0003720	Paxs	448.44	Joback Method
dvisc	0.0006468	Paxs	404.78	Joback Method
dvisc	0.0012854	Paxs	361.13	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1518532&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1518532&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/74-108-0/Benzene-1-3-bis-1-1-dimethylethyl-2-methoxy-5-methyl.pdf>

Generated by Cheméo on 2024-04-19 22:00:35.317910282 +0000 UTC m=+15853284.238487598.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.