

Cyclopropanecarbonitrile, 1,2-diphenyl-

Inchi:	InChI=1S/C16H13N/c17-12-16(14-9-5-2-6-10-14)11-15(16)13-7-3-1-4-8-13/h1-10,15H,11
InchiKey:	MIVVDTYTHSYGAF-UHFFFAOYSA-N
Formula:	C16H13N
SMILES:	N#CC1(c2ccccc2)CC1c1ccccc1
Mol. weight [g/mol]:	219.28
CAS:	10224-14-3

Physical Properties

Property code	Value	Unit	Source
gf	489.39	kJ/mol	Joback Method
hf	332.07	kJ/mol	Joback Method
hfus	19.69	kJ/mol	Joback Method
hvap	64.69	kJ/mol	Joback Method
ie	8.80 ± 0.08	eV	NIST Webbook
log10ws	-4.12		Crippen Method
logp	3.635		Crippen Method
mcvol	179.300	ml/mol	McGowan Method
pc	2563.69	kPa	Joback Method
tb	723.23	K	Joback Method
tc	991.75	K	Joback Method
tf	425.51	K	Joback Method
vc	0.696	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	485.40	J/mol×K	723.23	Joback Method
cpg	501.38	J/mol×K	767.98	Joback Method
cpg	516.52	J/mol×K	812.74	Joback Method
cpg	531.15	J/mol×K	857.49	Joback Method
cpg	545.58	J/mol×K	902.24	Joback Method
cpg	560.12	J/mol×K	946.99	Joback Method
cpg	575.10	J/mol×K	991.75	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C10224143&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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