

Glutaric acid, 2,2,3,3-tetrafluoropropyl 8-chlorooctyl ester

Inchi:	InChI=1S/C16H25ClF4O4/c17-10-5-3-1-2-4-6-11-24-13(22)8-7-9-14(23)25-12-16(20,21)
InchiKey:	SPMOJZIGRAGHQA-UHFFFAOYSA-N
Formula:	C16H25ClF4O4
SMILES:	O=C(CCCC(=O)OCC(F)(F)C(F)F)OCCCCCCCCCl
Mol. weight [g/mol]:	392.81

Physical Properties

Property code	Value	Unit	Source
gf	-1174.77	kJ/mol	Joback Method
hf	-1677.38	kJ/mol	Joback Method
hfus	48.35	kJ/mol	Joback Method
hvap	68.95	kJ/mol	Joback Method
log10ws	-5.03		Crippen Method
logp	4.723		Crippen Method
mcvol	270.500	ml/mol	McGowan Method
pc	1229.42	kPa	Joback Method
rinpol	2137.00		NIST Webbook
rinpol	2137.00		NIST Webbook
tb	748.90	K	Joback Method
tc	922.10	K	Joback Method
tf	434.10	K	Joback Method
vc	1.083	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	797.91	J/mol×K	748.90	Joback Method
cpg	812.40	J/mol×K	777.77	Joback Method
cpg	826.08	J/mol×K	806.63	Joback Method
cpg	838.96	J/mol×K	835.50	Joback Method
cpg	851.08	J/mol×K	864.37	Joback Method
cpg	862.45	J/mol×K	893.23	Joback Method
cpg	873.09	J/mol×K	922.10	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U393456&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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