

# 1,4,5,8-Naphthalenetetracarboxylic dianhydride

Other names:

Naphthalene-1,4,5,8-tetracarboxylic dianhydride  
[2]Benzopyrano[6,5,4-def][2]benzopyran-1,3,6,8-tetrone  
Naphthalenetetracarboxylic dianhydride  
1,4,5,8-Naphthalenetetracarboxylic acid anhydride  
1,4,5,8-Naphthalenetetracarboxylic acid dianhydride  
1,4,5,8-Naphthalenetetracarboxylic 1,8:4,5-dianhydride  
1,8:4,5-Naphthalenetetracarboxylic dianhydride  
2,7-Dioxapyrene-1,3,6,8-tetrone  
1,4,5,8-Naphthalenetetracarboxylic acid, 1,8:4,5-dianhydride  
NSC 84241  
NTCDA

naphthalene-1,8:4,5-tetracarboxylic dianhydride

Inchi: InChI=1S/C14H4O6/c15-11-5-1-2-6-10-8(14(18)20-12(6)16)4-3-7(9(5)10)13(17)19-11/h1

InchiKey: YTVNOVQHSGMMOV-UHFFFAOYSA-N

Formula: C14H4O6

SMILES: O=C1OC(=O)c2ccc3c4c(ccc1c24)C(=O)OC3=O

Mol. weight [g/mol]: 268.18

CAS: 81-30-1

## Physical Properties

Property code	Value	Unit	Source
gf	-278.14	kJ/mol	Joback Method
hf	-579.09	kJ/mol	Joback Method
hfus	29.65	kJ/mol	Joback Method
hvap	79.77	kJ/mol	Joback Method
log10ws	-4.19		Crippen Method
logp	1.461		Crippen Method
mcvol	161.200	ml/mol	McGowan Method
pc	3916.03	kPa	Joback Method
tb	933.30	K	Joback Method
tc	1222.26	K	Joback Method
tf	727.12	K	Joback Method
vc	0.620	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	489.48	J/mol×K	933.30	Joback Method
cpg	498.71	J/mol×K	981.46	Joback Method
cpg	506.37	J/mol×K	1029.62	Joback Method
cpg	512.43	J/mol×K	1077.78	Joback Method
cpg	516.85	J/mol×K	1125.94	Joback Method
cpg	519.60	J/mol×K	1174.10	Joback Method
cpg	520.64	J/mol×K	1222.26	Joback Method

## Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C81301&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C81301&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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