

**Fumaric acid,
di(2,2,3,3,4,4,5,5-octafluoropentyl) ester**

Inchi: InChI=1S/C14H8F16O4/c15-7(16)11(23,24)13(27,28)9(19,20)3-33-5(31)1-2-6(32)34-4-10

InchiKey: GJKYQRGXEQSQOI-OWOJBTEDSA-N

Formula: C₁₄H₈F₁₆O₄

SMILES: O=C(C=CC(=O)OCC(F)(F)C(F)(F)C(F)(F)C(F)F)OCC(F)(F)C(F)(F)C(F)(F)C(F)F

Mol. weight [g/mol]: 544.19

Physical Properties

Property code	Value	Unit	Source
gf	-3425.42	kJ/mol	Joback Method
hf	-3905.49	kJ/mol	Joback Method
hfus	35.54	kJ/mol	Joback Method
hvap	43.40	kJ/mol	Joback Method
log10ws	-5.77		Crippen Method
logp	4.971		Crippen Method
mcvol	247.020	ml/mol	McGowan Method
pc	1105.21	kPa	Joback Method
rinpol	1434.00		NIST Webbook
rinpol	1434.00		NIST Webbook
tb	644.52	K	Joback Method
tc	792.42	K	Joback Method
tf	380.74	K	Joback Method
vc	1.058	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	745.65	J/mol×K	644.52	Joback Method
cpg	756.78	J/mol×K	669.17	Joback Method
cpg	767.09	J/mol×K	693.82	Joback Method
cpg	776.65	J/mol×K	718.47	Joback Method
cpg	785.50	J/mol×K	743.12	Joback Method
cpg	793.71	J/mol×K	767.77	Joback Method
cpg	801.31	J/mol×K	792.42	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U348797&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rlnol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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