

# Benzenamine, N-ethyl-N-phenyl-

<b>Other names:</b>	N,N-diphenyl-N-ethylamine
<b>Inchi:</b>	InChI=1S/C14H15N/c1-2-15(13-9-5-3-6-10-13)14-11-7-4-8-12-14/h3-12H,2H2,1H3
<b>InchiKey:</b>	ITMSSZATZARZCA-UHFFFAOYSA-N
<b>Formula:</b>	C14H15N
<b>SMILES:</b>	CCN(c1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	197.28
<b>CAS:</b>	606-99-5

## Physical Properties

Property code	Value	Unit	Source
gf	402.60	kJ/mol	Joback Method
hf	208.30	kJ/mol	Joback Method
hfus	23.12	kJ/mol	Joback Method
hvap	53.35	kJ/mol	Joback Method
log10ws	-3.59		Crippen Method
logp	3.845		Crippen Method
mcvol	170.580	ml/mol	McGowan Method
pc	2764.26	kPa	Joback Method
tb	585.52	K	Joback Method
tc	821.07	K	Joback Method
tf	332.85	K	Joback Method
vc	0.622	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	408.79	J/molxK	585.52	Joback Method
cpg	426.97	J/molxK	624.78	Joback Method
cpg	443.75	J/molxK	664.04	Joback Method
cpg	459.20	J/molxK	703.29	Joback Method
cpg	473.44	J/molxK	742.55	Joback Method
cpg	486.53	J/molxK	781.81	Joback Method
cpg	498.57	J/molxK	821.07	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C606995&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C606995&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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