

# 2,9b-Dihydrofurano[4,3,2-jk]fluorene

<b>Other names:</b>	Dihydrofurano[4,3,2-jk]fluorene
<b>Inchi:</b>	InChI=1S/C14H10O/c1-2-6-12-10(5-1)11-7-3-4-9-8-15-14(12)13(9)11/h1-7,14H,8H2
<b>InchiKey:</b>	HQEUANYBCCGWMQ-UHFFFAOYSA-N
<b>Formula:</b>	C14H10O
<b>SMILES:</b>	<chem>c1ccc2c(c1)-c1cccc3c1C2OC3</chem>
<b>Mol. weight [g/mol]:</b>	194.23
<b>CAS:</b>	204396-15-6

## Physical Properties

Property code	Value	Unit	Source
gf	342.32	kJ/mol	Joback Method
hf	158.78	kJ/mol	Joback Method
hfus	28.41	kJ/mol	Joback Method
hvap	57.42	kJ/mol	Joback Method
log10ws	-4.74		Crippen Method
logp	3.287		Crippen Method
mcvol	144.750	ml/mol	McGowan Method
pc	3318.18	kPa	Joback Method
rinpol	274.10		NIST Webbook
rinpol	274.10		NIST Webbook
tb	620.31	K	Joback Method
tc	869.54	K	Joback Method
tf	415.19	K	Joback Method
vc	0.564	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	365.32	J/mol×K	620.31	Joback Method
cpg	423.59	J/mol×K	828.00	Joback Method
cpg	413.58	J/mol×K	786.46	Joback Method
cpg	402.93	J/mol×K	744.92	Joback Method
cpg	391.47	J/mol×K	703.39	Joback Method
cpg	378.99	J/mol×K	661.85	Joback Method

cpg	433.15	J/molxK	869.54	Joback Method
dvisc	0.0019664	Paxs	620.31	Joback Method
dvisc	0.0020311	Paxs	586.12	Joback Method
dvisc	0.0021064	Paxs	551.94	Joback Method
dvisc	0.0021951	Paxs	517.75	Joback Method
dvisc	0.0023008	Paxs	483.56	Joback Method
dvisc	0.0024289	Paxs	449.38	Joback Method
dvisc	0.0025872	Paxs	415.19	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C204396156&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C204396156&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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