

# Ethyldicyclohexylamine

<b>Other names:</b>	N-Ethyldicyclohexylamine Cyclohexanamine, N-cyclohexyl-N-ethyl- Dicyclohexylamine, N-ethyl- N-cyclohexyl-N-ethylcyclohexylamine
<b>Inchi:</b>	InChI=1S/C14H27N/c1-2-15(13-9-5-3-6-10-13)14-11-7-4-8-12-14/h13-14H,2-12H2,1H3
<b>InchiKey:</b>	XRKQMIFKHDXFNQ-UHFFFAOYSA-N
<b>Formula:</b>	C14H27N
<b>SMILES:</b>	CCN(C1CCCCC1)C1CCCCC1
<b>Mol. weight [g/mol]:</b>	209.37
<b>CAS:</b>	7175-49-7

## Physical Properties

Property code	Value	Unit	Source
gf	226.68	kJ/mol	Joback Method
hf	-156.12	kJ/mol	Joback Method
hfus	18.71	kJ/mol	Joback Method
hvap	49.66	kJ/mol	Joback Method
log10ws	-4.27		Crippen Method
logp	3.974		Crippen Method
mcvol	196.380	ml/mol	McGowan Method
pc	2155.30	kPa	Joback Method
tb	571.26	K	Joback Method
tc	791.42	K	Joback Method
tf	294.77	K	Joback Method
vc	0.704	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	529.64	J/molxK	571.26	Joback Method
cpg	556.40	J/molxK	607.95	Joback Method
cpg	581.47	J/molxK	644.65	Joback Method
cpg	604.92	J/molxK	681.34	Joback Method
cpg	626.81	J/molxK	718.03	Joback Method

cpg	647.21	J/mol×K	754.72	Joback Method
cpg	666.17	J/mol×K	791.42	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	387.70	K	0.40	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7175497&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7175497&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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