

1-((1S,3aR,4S,7R,7aR)-3a,7a-Dimethyloctahydro-1-

Inchi:	InChI=1S/C14H22O/c1-9(15)12-6-7-13(2)10-4-5-11(8-10)14(12,13)3/h10-12H,4-8H2,1-3H
InchiKey:	ZVJYJEILWCYOQE-UHFFFAOYSA-N
Formula:	C14H22O
SMILES:	CC(=O)C1CCC2(C)C3CCC(C3)C12C
Mol. weight [g/mol]:	206.32
CAS:	66748-84-3

Physical Properties

Property code	Value	Unit	Source
gf	81.83	kJ/mol	Joback Method
hf	-242.83	kJ/mol	Joback Method
hfus	15.47	kJ/mol	Joback Method
hvap	50.50	kJ/mol	Joback Method
log10ws	-3.44		Crippen Method
logp	3.428		Crippen Method
mcvol	177.110	ml/mol	McGowan Method
pc	2356.49	kPa	Joback Method
rinpol	1589.40		NIST Webbook
tb	589.22	K	Joback Method
tc	815.60	K	Joback Method
tf	387.09	K	Joback Method
vc	0.682	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	497.75	J/molxK	589.22	Joback Method
cpg	518.74	J/molxK	626.95	Joback Method
cpg	538.31	J/molxK	664.68	Joback Method
cpg	556.79	J/molxK	702.41	Joback Method
cpg	574.48	J/molxK	740.14	Joback Method
cpg	591.69	J/molxK	777.87	Joback Method
cpg	608.72	J/molxK	815.60	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C66748843&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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