

# Benzene, 1,2,4,5-tetraethyl-

<b>Other names:</b>	1,2,4,5-Tetraethylbenzene
<b>Inchi:</b>	InChI=1S/C14H22/c1-5-11-9-13(7-3)14(8-4)10-12(11)6-2/h9-10H,5-8H2,1-4H3
<b>InchiKey:</b>	NSOYZSQQDPNCAC-UHFFFAOYSA-N
<b>Formula:</b>	C14H22
<b>SMILES:</b>	CCc1cc(CC)c(CC)cc1CC
<b>Mol. weight [g/mol]:</b>	190.32
<b>CAS:</b>	635-81-4

## Physical Properties

Property code	Value	Unit	Source
gf	150.52	kJ/mol	Joback Method
hf	-130.17	kJ/mol	Joback Method
hfus	24.89	kJ/mol	Joback Method
hvap	51.02	kJ/mol	Joback Method
log10ws	-4.68		Crippen Method
logp	3.936		Crippen Method
mcvol	184.360	ml/mol	McGowan Method
pc	1930.44	kPa	Joback Method
rinpol	1388.00		NIST Webbook
tb	561.34	K	Joback Method
tc	758.43	K	Joback Method
tf	283.00 ± 2.00	K	NIST Webbook
tf	283.00 ± 2.00	K	NIST Webbook
vc	0.712	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	535.27	J/molxK	758.43	Joback Method
cpg	521.83	J/molxK	725.58	Joback Method
cpg	507.66	J/molxK	692.73	Joback Method
cpg	492.74	J/molxK	659.89	Joback Method
cpg	477.05	J/molxK	627.04	Joback Method
cpg	460.57	J/molxK	594.19	Joback Method

cpg	443.27	J/molxK	561.34	Joback Method
dvisc	0.0014453	Paxs	311.52	Joback Method
dvisc	0.0001675	Paxs	561.34	Joback Method
dvisc	0.0002078	Paxs	519.70	Joback Method
dvisc	0.0002676	Paxs	478.07	Joback Method
dvisc	0.0003615	Paxs	436.43	Joback Method
dvisc	0.0005205	Paxs	394.79	Joback Method
dvisc	0.0008167	Paxs	353.16	Joback Method
hvapt	54.50	kJ/mol	429.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.42990e+01
Coeff. B	-4.63058e+03
Coeff. C	-5.56700e+01
Temperature range (K), min.	386.16
Temperature range (K), max.	570.90

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol710.mol">https://www.thermo.com/files/research/kdb/mol/mol710.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C635814&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C635814&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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