

N-phenethyl-2-norbornanamine

Inchi:	InChI=1S/C15H21N/c1-2-4-12(5-3-1)8-9-16-15-11-13-6-7-14(15)10-13/h1-5,13-16H,6-11
InchiKey:	UOJWJKSWEPSVQD-UHFFFAOYSA-N
Formula:	C15H21N
SMILES:	<chem>c1ccc(CCNC2CC3CCC2C3)cc1</chem>
Mol. weight [g/mol]:	215.33
CAS:	4935-80-2

Physical Properties

Property code	Value	Unit	Source
gf	378.91	kJ/mol	Joback Method
hf	56.17	kJ/mol	Joback Method
hfus	28.99	kJ/mol	Joback Method
hvap	57.38	kJ/mol	Joback Method
log10ws	-3.81		Crippen Method
logp	3.007		Crippen Method
mcvol	186.710	ml/mol	McGowan Method
pc	2351.92	kPa	Joback Method
tb	632.53	K	Joback Method
tc	858.78	K	Joback Method
tf	366.01	K	Joback Method
vc	0.708	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	527.24	J/molxK	632.53	Joback Method
cpg	548.07	J/molxK	670.24	Joback Method
cpg	567.42	J/molxK	707.95	Joback Method
cpg	585.41	J/molxK	745.65	Joback Method
cpg	602.14	J/molxK	783.36	Joback Method
cpg	617.73	J/molxK	821.07	Joback Method
cpg	632.28	J/molxK	858.78	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C4935802&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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