

# 4-tert-Butyl-1(1-thioxo-2,2-dimethyl-propyl)-benzene

<b>Other names:</b>	1-Propanethione, 1-[4-(1,1-dimethylethyl)phenyl]-2,2-dimethyl-
<b>Inchi:</b>	InChI=1S/C15H22S/c1-14(2,3)12-9-7-11(8-10-12)13(16)15(4,5)6/h7-10H,1-6H3
<b>InchiKey:</b>	ZPRFQKNKXHCDCW-UHFFFAOYSA-N
<b>Formula:</b>	C15H22S
<b>SMILES:</b>	CC(C)(C)C(=S)c1ccc(C(C)(C)C)cc1
<b>Mol. weight [g/mol]:</b>	234.40
<b>CAS:</b>	72194-24-2

## Physical Properties

Property code	Value	Unit	Source
gf	300.94	kJ/mol	Joback Method
hf	1.13	kJ/mol	Joback Method
hfus	18.03	kJ/mol	Joback Method
hvap	56.06	kJ/mol	Joback Method
ie	7.85	eV	NIST Webbook
ie	8.96	eV	NIST Webbook
log10ws	-5.06		Crippen Method
logp	4.748		Crippen Method
mcvol	210.500	ml/mol	McGowan Method
pc	2043.76	kPa	Joback Method
tb	637.84	K	Joback Method
tc	877.23	K	Joback Method
tf	336.86	K	Joback Method
vc	0.781	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	540.24	J/molxK	637.84	Joback Method
cpg	558.89	J/molxK	677.74	Joback Method
cpg	576.07	J/molxK	717.64	Joback Method
cpg	591.93	J/molxK	757.54	Joback Method
cpg	606.65	J/molxK	797.43	Joback Method
cpg	620.39	J/molxK	837.33	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C72194242&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C72194242&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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