

# 9-Decen-1-ol, methyl ether

Inchi:	InChI=1S/C11H22O/c1-3-4-5-6-7-8-9-10-11-12-2/h3H,1,4-11H2,2H3
InchiKey:	UIGGBQKGVMIQC-UHFFFAOYSA-N
Formula:	C11H22O
SMILES:	C=CCCCCCCCOC
Mol. weight [g/mol]:	170.29

## Physical Properties

Property code	Value	Unit	Source
gf	24.58	kJ/mol	Joback Method
hf	-277.16	kJ/mol	Joback Method
hfus	24.15	kJ/mol	Joback Method
hvap	41.82	kJ/mol	Joback Method
log10ws	-3.37		Crippen Method
logp	3.550		Crippen Method
mvol	167.420	ml/mol	McGowan Method
pc	1977.07	kPa	Joback Method
rinpol	1229.90		NIST Webbook
rinpol	1229.90		NIST Webbook
tb	470.18	K	Joback Method
tc	634.93	K	Joback Method
tf	234.20	K	Joback Method
vc	0.650	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	370.42	J/molxK	470.18	Joback Method
cpg	385.59	J/molxK	497.64	Joback Method
cpg	400.22	J/molxK	525.10	Joback Method
cpg	414.30	J/molxK	552.56	Joback Method
cpg	427.86	J/molxK	580.01	Joback Method
cpg	440.91	J/molxK	607.47	Joback Method
cpg	453.44	J/molxK	634.93	Joback Method
dvisc	0.0038824	Paxs	234.20	Joback Method

dvisc	0.0016405	Paxs	273.53	Joback Method
dvisc	0.0008608	Paxs	312.86	Joback Method
dvisc	0.0005217	Paxs	352.19	Joback Method
dvisc	0.0003496	Paxs	391.52	Joback Method
dvisc	0.0002521	Paxs	430.85	Joback Method
dvisc	0.0001920	Paxs	470.18	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U352651&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U352651&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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