

3-Decanone, 2-methyl-

Other names:	2-Methyl-3-decanone 2-methyldecan-3-one
Inchi:	InChI=1S/C11H22O/c1-4-5-6-7-8-9-11(12)10(2)3/h10H,4-9H2,1-3H3
InchiKey:	HJSFZKMTFMWEAQ-UHFFFAOYSA-N
Formula:	C11H22O
SMILES:	CCCCCCCC(=O)C(C)C
Mol. weight [g/mol]:	170.29
CAS:	5463-82-1

Physical Properties

Property code	Value	Unit	Source
gf	-89.62	kJ/mol	Joback Method
hf	-388.23	kJ/mol	Joback Method
hfus	22.32	kJ/mol	Joback Method
hvap	46.44	kJ/mol	Joback Method
log10ws	-3.47		Crippen Method
logp	3.572		Crippen Method
mcvol	167.420	ml/mol	McGowan Method
pc	2060.49	kPa	Joback Method
tb	504.51	K	Joback Method
tc	679.38	K	Joback Method
tf	248.66	K	Joback Method
vc	0.651	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	388.41	J/mol×K	504.51	Joback Method
cpg	404.02	J/mol×K	533.65	Joback Method
cpg	418.97	J/mol×K	562.80	Joback Method
cpg	433.29	J/mol×K	591.94	Joback Method
cpg	446.99	J/mol×K	621.09	Joback Method
cpg	460.10	J/mol×K	650.23	Joback Method
cpg	472.61	J/mol×K	679.38	Joback Method

dvisc	0.0065882	Pa×s	248.66	Joback Method
dvisc	0.0025234	Pa×s	291.30	Joback Method
dvisc	0.0012349	Pa×s	333.94	Joback Method
dvisc	0.0007105	Pa×s	376.59	Joback Method
dvisc	0.0004574	Pa×s	419.23	Joback Method
dvisc	0.0003195	Pa×s	461.87	Joback Method
dvisc	0.0002371	Pa×s	504.51	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.45783e+01
Coeff. B	-4.18822e+03
Coeff. C	-7.76060e+01
Temperature range (K), min.	370.68
Temperature range (K), max.	529.57

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C5463821&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀w_s:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
p_c:	Critical Pressure
p_{vap}:	Vapor pressure
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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