

# 1-(4-Benzylphenyl)-1-butanone

<b>Inchi:</b>	InChI=1S/C17H18O/c1-2-6-17(18)16-11-9-15(10-12-16)13-14-7-4-3-5-8-14/h3-5,7-12H,2
<b>InchiKey:</b>	CQGHAVUWNSCNMH-UHFFFAOYSA-N
<b>Formula:</b>	C17H18O
<b>SMILES:</b>	CCCC(=O)c1ccc(Cc2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	238.32
<b>CAS:</b>	17302-69-1

## Physical Properties

Property code	Value	Unit	Source
gf	178.53	kJ/mol	Joback Method
hf	-45.20	kJ/mol	Joback Method
hfus	29.08	kJ/mol	Joback Method
hvap	65.40	kJ/mol	Joback Method
log10ws	-5.07		Crippen Method
logp	4.260		Crippen Method
mcvol	204.440	ml/mol	McGowan Method
pc	2183.60	kPa	Joback Method
tb	700.57	K	Joback Method
tc	933.15	K	Joback Method
tf	396.64	K	Joback Method
vc	0.777	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	544.35	J/molxK	700.57	Joback Method
cpg	561.15	J/molxK	739.33	Joback Method
cpg	576.71	J/molxK	778.10	Joback Method
cpg	591.09	J/molxK	816.86	Joback Method
cpg	604.36	J/molxK	855.62	Joback Method
cpg	616.60	J/molxK	894.39	Joback Method
cpg	627.89	J/molxK	933.15	Joback Method
dvisc	0.0015064	Paxs	396.64	Joback Method
dvisc	0.0008101	Paxs	447.29	Joback Method

dvisc	0.0004942	Paxs	497.95	Joback Method
dvisc	0.0003303	Paxs	548.61	Joback Method
dvisc	0.0002363	Paxs	599.26	Joback Method
dvisc	0.0001781	Paxs	649.91	Joback Method
dvisc	0.0001399	Paxs	700.57	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C17302691&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C17302691&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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