

# 1,4-Pentadien-3-one, 1,5-diphenyl-, (Z,Z)-

<b>Other names:</b>	(1Z,4Z)-1,5-Diphenyl-1,4-pentadien-3-one
<b>Inchi:</b>	InChI=1S/C17H14O/c18-17(13-11-15-7-3-1-4-8-15)14-12-16-9-5-2-6-10-16/h1-14H/b13-
<b>InchiKey:</b>	WMKGGPCROCCUDY-XSYHWHKQSA-N
<b>Formula:</b>	C17H14O
<b>SMILES:</b>	O=C(C=Cc1ccccc1)C=Cc1ccccc1
<b>Mol. weight [g/mol]:</b>	234.29
<b>CAS:</b>	58321-78-1

## Physical Properties

Property code	Value	Unit	Source
gf	348.60	kJ/mol	Joback Method
hf	200.71	kJ/mol	Joback Method
hfus	29.87	kJ/mol	Joback Method
hvap	64.65	kJ/mol	Joback Method
log10ws	-4.47		Crippen Method
logp	3.982		Crippen Method
mcvol	195.840	ml/mol	McGowan Method
pc	2448.32	kPa	Joback Method
rinpol	2122.00		NIST Webbook
rinpol	2122.00		NIST Webbook
tb	703.91	K	Joback Method
tc	954.96	K	Joback Method
tf	373.96	K	Joback Method
vc	0.738	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	498.70	J/molxK	703.91	Joback Method
cpg	514.49	J/molxK	745.75	Joback Method
cpg	528.96	J/molxK	787.59	Joback Method
cpg	542.24	J/molxK	829.43	Joback Method
cpg	554.49	J/molxK	871.27	Joback Method
cpg	565.84	J/molxK	913.11	Joback Method

cpg	576.44	J/molxK	954.96	Joback Method
dvisc	0.0015983	Paxs	373.96	Joback Method
dvisc	0.0007579	Paxs	428.95	Joback Method
dvisc	0.0004258	Paxs	483.94	Joback Method
dvisc	0.0002691	Paxs	538.93	Joback Method
dvisc	0.0001851	Paxs	593.93	Joback Method
dvisc	0.0001357	Paxs	648.92	Joback Method
dvisc	0.0001044	Paxs	703.91	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C58321781&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C58321781&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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