

# 7-Heptadecanone

<b>Inchi:</b>	InChI=1S/C17H34O/c1-3-5-7-9-10-11-12-14-16-17(18)15-13-8-6-4-2/h3-16H2,1-2H3
<b>InchiKey:</b>	NFRKSAMCQGIGRC-UHFFFAOYSA-N
<b>Formula:</b>	C17H34O
<b>SMILES:</b>	CCCCCCCCCCC(=O)CCCCC
<b>Mol. weight [g/mol]:</b>	254.45
<b>CAS:</b>	6064-42-2

## Physical Properties

Property code	Value	Unit	Source
gf	-36.66	kJ/mol	Joback Method
hf	-506.79	kJ/mol	Joback Method
hfus	41.38	kJ/mol	Joback Method
hvap	60.18	kJ/mol	Joback Method
log10ws	-6.22		Crippen Method
logp	6.057		Crippen Method
mvol	251.960	ml/mol	McGowan Method
pc	1287.44	kPa	Joback Method
tb	642.23	K	Joback Method
tc	809.06	K	Joback Method
tf	331.28	K	Joback Method
vc	0.994	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	700.79	J/molxK	642.23	Joback Method
cpg	719.37	J/molxK	670.04	Joback Method
cpg	737.15	J/molxK	697.84	Joback Method
cpg	754.16	J/molxK	725.65	Joback Method
cpg	770.42	J/molxK	753.45	Joback Method
cpg	785.94	J/molxK	781.26	Joback Method
cpg	800.76	J/molxK	809.06	Joback Method
dvisc	0.0032643	Paxs	331.28	Joback Method
dvisc	0.0013527	Paxs	383.11	Joback Method

dvisc	0.0006915	Paxs	434.93	Joback Method
dvisc	0.0004078	Paxs	486.75	Joback Method
dvisc	0.0002662	Paxs	538.58	Joback Method
dvisc	0.0001873	Paxs	590.40	Joback Method
dvisc	0.0001395	Paxs	642.23	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.47776e+01
Coeff. B	-4.98789e+03
Coeff. C	-1.04850e+02
Temperature range (K), min.	449.08
Temperature range (K), max.	631.77

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6064422&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6064422&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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