

# Cyclohexane, 1,1'-(1,5-pentanediy)bis-

Inchi:	InChI=1S/C17H32/c1-4-10-16(11-5-1)14-8-3-9-15-17-12-6-2-7-13-17/h16-17H,1-15H2
InchiKey:	ALPSDJDPJHLYKS-UHFFFAOYSA-N
Formula:	C17H32
SMILES:	C1CCC(CCCCC2CCCCC2)CC1
Mol. weight [g/mol]:	236.44
CAS:	54833-31-7

## Physical Properties

Property code	Value	Unit	Source
chl	-10930.00	kJ/mol	NIST Webbook
gf	141.16	kJ/mol	Joback Method
hf	-285.57	kJ/mol	Joback Method
hfus	23.46	kJ/mol	Joback Method
hvap	54.29	kJ/mol	Joback Method
log10ws	-6.24		Crippen Method
logp	6.098		Crippen Method
mcvol	228.670	ml/mol	McGowan Method
pc	1679.66	kPa	Joback Method
tb	584.00 ± 5.00	K	NIST Webbook
tb	598.30 ± 0.30	K	NIST Webbook
tb	584.00 ± 5.00	K	NIST Webbook
tb	588.00 ± 8.00	K	NIST Webbook
tc	839.61	K	Joback Method
tf	259.54 ± 1.00	K	NIST Webbook
tf	259.54 ± 0.20	K	NIST Webbook
vc	0.854	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	658.62	J/mol×K	627.46	Joback Method
cpg	776.38	J/mol×K	804.25	Joback Method
cpg	755.83	J/mol×K	768.89	Joback Method
cpg	733.83	J/mol×K	733.54	Joback Method

cpg	710.34	J/molxK	698.18	Joback Method
cpg	685.29	J/molxK	662.82	Joback Method
cpg	795.54	J/molxK	839.61	Joback Method
dvisc	0.0001557	Paxs	627.46	Joback Method
dvisc	0.0002171	Paxs	572.24	Joback Method
dvisc	0.0003249	Paxs	517.01	Joback Method
dvisc	0.0005354	Paxs	461.79	Joback Method
dvisc	0.0010106	Paxs	406.56	Joback Method
dvisc	0.0023293	Paxs	351.34	Joback Method
dvisc	0.0073304	Paxs	296.11	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C54833317&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C54833317&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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