

# Tetrahydroaplotaxene

**Inchi:** InChI=1S/C17H32/c1-3-5-7-9-11-13-15-17-16-14-12-10-8-6-4-2/h3,15,17H,1,4-14,16H2,2  
**InchiKey:** ZCNSOBXQEHNQMJ-ICFOKQHNSA-N  
**Formula:** C17H32  
**SMILES:** C=CCCCCCC=CCCCCCCCC  
**Mol. weight [g/mol]:** 236.44

## Physical Properties

Property code	Value	Unit	Source
gf	260.32	kJ/mol	Joback Method
hf	-151.56	kJ/mol	Joback Method
hfus	38.71	kJ/mol	Joback Method
hvap	52.72	kJ/mol	Joback Method
log10ws	-6.65		Crippen Method
logp	6.430		Crippen Method
mvol	241.790	ml/mol	McGowan Method
pc	1316.56	kPa	Joback Method
ripol	1758.00		NIST Webbook
ripol	1758.00		NIST Webbook
tb	589.20	K	Joback Method
tc	755.29	K	Joback Method
tf	274.51	K	Joback Method
vc	0.949	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	626.58	J/molxK	589.20	Joback Method
cpg	645.48	J/molxK	616.88	Joback Method
cpg	663.56	J/molxK	644.56	Joback Method
cpg	680.85	J/molxK	672.25	Joback Method
cpg	697.39	J/molxK	699.93	Joback Method
cpg	713.21	J/molxK	727.61	Joback Method
cpg	728.34	J/molxK	755.29	Joback Method
dvisc	0.0042107	Paxs	274.51	Joback Method

dvisc	0.0014518	Paxs	326.96	Joback Method
dvisc	0.0006719	Paxs	379.41	Joback Method
dvisc	0.0003749	Paxs	431.86	Joback Method
dvisc	0.0002374	Paxs	484.30	Joback Method
dvisc	0.0001644	Paxs	536.75	Joback Method
dvisc	0.0001215	Paxs	589.20	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R337185&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R337185&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/74-531-0/Tetrahydroaplotaxene.pdf>

Generated by Cheméo on 2025-12-05 08:24:49.660235203 +0000 UTC m=+4671287.190275858.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.