

# Bicyclo[2.2.1]hept-5-ene-2,3-dicarboxylic acid, 1,4,5,6,7,7-hexachloro-, dibutyl ester

Other names:

Chlorendic acid dibutyl ester

Dibutyl chlorendate

5-Norbornene-2,3-dicarboxylic acid, 1,4,5,6,7,7-hexachloro-, dibutyl ester

dibutyl 1,4,5,6,7,7-hexachlorobicyclo[2.2.1]hept-5-ene-2,3-dicarboxylate

**Inchi:** InChI=1S/C17H20Cl6O4/c1-3-5-7-26-13(24)9-10(14(25)27-8-6-4-2)16(21)12(19)11(18)13

**InchiKey:** UJAHPBDUQZFDLA-UHFFFAOYSA-N

**Formula:** C17H20Cl6O4

**SMILES:** CCCCOC(=O)C1C(C(=O)OCCCC)C2(Cl)C(Cl)=C(Cl)C1(Cl)C2(Cl)Cl

**Mol. weight [g/mol]:** 501.06

**CAS:** 1770-80-5

## Physical Properties

Property code	Value	Unit	Source
gf	-366.66	kJ/mol	Joback Method
hf	-819.27	kJ/mol	Joback Method
hfus	49.48	kJ/mol	Joback Method
hvap	95.29	kJ/mol	Joback Method
log10ws	-6.56		Crippen Method
logp	5.751		Crippen Method
mcvol	312.690	ml/mol	McGowan Method
pc	1420.78	kPa	Joback Method
tb	979.10	K	Joback Method
tc	1214.37	K	Joback Method
tf	722.33	K	Joback Method
vc	1.212	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	913.49	J/molxK	979.10	Joback Method
cpg	939.11	J/molxK	1018.31	Joback Method
cpg	967.08	J/molxK	1057.52	Joback Method
cpg	997.79	J/molxK	1096.74	Joback Method
cpg	1031.60	J/molxK	1135.95	Joback Method

cpg	1068.91	J/mol×K	1175.16	Joback Method
cpg	1110.08	J/mol×K	1214.37	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1770805&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1770805&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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