

# 2-(2-[(9Z)-9-octadecenoyloxy]ethoxy)ethyl (9z)-9-octadecenoate

Other names:	oxydi-2,1-ethanediyl dioleate
Inchi:	InChI=1S/C40H74O5/c1-3-5-7-9-11-13-15-17-19-21-23-25-27-29-31-33-39(41)44-37-35-
InchiKey:	HJRDNARELSKHEF-CLFAGFIQSA-N
Formula:	C40H74O5
SMILES:	CCCCCCCCC=CCCCCCCCC(=O)OCCOCCOC(=O)CCCCCCCCC=CCCCCCCCC
Mol. weight [g/mol]:	635.01
CAS:	21209-30-3

## Physical Properties

Property code	Value	Unit	Source
gf	-126.48	kJ/mol	Joback Method
hf	-1256.31	kJ/mol	Joback Method
hfus	106.52	kJ/mol	Joback Method
hvap	125.27	kJ/mol	Joback Method
log10ws	-13.09		Crippen Method
logp	12.164		Crippen Method
mcvol	586.610	ml/mol	McGowan Method
pc	420.09	kPa	Joback Method
tb	1297.92	K	Joback Method
tc	1830.91	K	Joback Method
tf	696.95	K	Joback Method
vc	2.301	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	2240.22	J/molxK	1297.92	Joback Method
cpg	2276.11	J/molxK	1386.75	Joback Method
cpg	2306.00	J/molxK	1475.58	Joback Method
cpg	2331.47	J/molxK	1564.41	Joback Method
cpg	2354.10	J/molxK	1653.25	Joback Method
cpg	2375.49	J/molxK	1742.08	Joback Method
cpg	2397.22	J/molxK	1830.91	Joback Method
dvisc	0.0000315	Paxs	696.95	Joback Method

dvisc	0.0000130	Paxs	797.11	Joback Method
dvisc	0.0000066	Paxs	897.27	Joback Method
dvisc	0.0000038	Paxs	997.43	Joback Method
dvisc	0.0000024	Paxs	1097.60	Joback Method
dvisc	0.0000017	Paxs	1197.76	Joback Method
dvisc	0.0000012	Paxs	1297.92	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C21209303&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C21209303&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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