

Hexaphenylbenzene

Other names:	Benzene, hexaphenyl- 1,1':2',1''-Terphenyl, 3',4',5',6'-tetraphenyl- m-Terphenyl, 2',4',5',6'-tetraphenyl- 3',4',5',6'-tetraphenyl-o-terphenyl
Inchi:	InChI=1S/C42H30/c1-7-19-31(20-8-1)37-38(32-21-9-2-10-22-32)40(34-25-13-4-14-26-34)
InchiKey:	QBHWPVJPWQGYDS-UHFFFAOYSA-N
Formula:	C42H30
SMILES:	<chem>c1ccc(-c2c(-c3ccccc3)c(-c3ccccc3)c(-c3ccccc3)c(-c3ccccc3)c2-c2ccccc2)cc1</chem>
Mol. weight [g/mol]:	534.69
CAS:	992-04-1

Physical Properties

Property code	Value	Unit	Source
gf	1041.48	kJ/mol	Joback Method
hf	688.15	kJ/mol	Joback Method
hfus	60.88	kJ/mol	Joback Method
hvap	128.33	kJ/mol	Joback Method
ie	8.47 ± 0.05	eV	NIST Webbook
log10ws	-17.09		Crippen Method
logp	11.689		Crippen Method
mcvol	436.320	ml/mol	McGowan Method
pc	1105.21	kPa	Joback Method
tb	1372.02	K	Joback Method
tc	1688.44	K	Joback Method
tf	810.64	K	Joback Method
vc	1.631	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1477.57	J/mol×K	1372.02	Joback Method
cpg	1579.28	J/mol×K	1635.71	Joback Method
cpg	1555.08	J/mol×K	1582.97	Joback Method
cpg	1533.29	J/mol×K	1530.23	Joback Method

cpg	1513.42	J/molxK	1477.49	Joback Method
cpg	1495.01	J/molxK	1424.76	Joback Method
cpg	1606.36	J/molxK	1688.44	Joback Method
dvisc	0.0000065	Paxs	1372.02	Joback Method
dvisc	0.0000081	Paxs	1278.46	Joback Method
dvisc	0.0000106	Paxs	1184.89	Joback Method
dvisc	0.0000144	Paxs	1091.33	Joback Method
dvisc	0.0000207	Paxs	997.77	Joback Method
dvisc	0.0000322	Paxs	904.20	Joback Method
dvisc	0.0000553	Paxs	810.64	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C992041&Units=SI

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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