

# Docosyl heptyl ether

**Inchi:** InChI=1S/C29H60O/c1-3-5-7-9-10-11-12-13-14-15-16-17-18-19-20-21-22-23-25-27-29-31  
**InchiKey:** LIMCJTPrXZHVLO-UHFFFAOYSA-N  
**Formula:** C<sub>29</sub>H<sub>60</sub>O  
**SMILES:** CCCCCCCCCCCCCCCCCCCCCCOCCCCCCC  
**Mol. weight [g/mol]:** 424.79

## Physical Properties

Property code	Value	Unit	Source
gf	88.30	kJ/mol	Joback Method
hf	-774.11	kJ/mol	Joback Method
hfus	72.05	kJ/mol	Joback Method
hvap	82.56	kJ/mol	Joback Method
log10ws	-11.05		Crippen Method
logp	10.795		Crippen Method
mvol	425.340	ml/mol	McGowan Method
pc	619.10	kPa	Joback Method
rinpol	2966.00		NIST Webbook
rinpol	2966.00		NIST Webbook
tb	885.34	K	Joback Method
tc	1090.98	K	Joback Method
tf	438.82	K	Joback Method
vc	1.677	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1451.36	J/mol×K	885.34	Joback Method
cpg	1477.87	J/mol×K	919.61	Joback Method
cpg	1502.80	J/mol×K	953.89	Joback Method
cpg	1526.22	J/mol×K	988.16	Joback Method
cpg	1548.19	J/mol×K	1022.44	Joback Method
cpg	1568.78	J/mol×K	1056.71	Joback Method
cpg	1588.06	J/mol×K	1090.98	Joback Method
dvisc	0.0007935	Paxs	438.82	Joback Method

dvisc	0.0002790	Paxs	513.24	Joback Method
dvisc	0.0001278	Paxs	587.66	Joback Method
dvisc	0.0000698	Paxs	662.08	Joback Method
dvisc	0.0000431	Paxs	736.50	Joback Method
dvisc	0.0000290	Paxs	810.92	Joback Method
dvisc	0.0000209	Paxs	885.34	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406397&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406397&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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