

# 1-Nonacosene

<b>Inchi:</b>	InChI=1S/C29H58/c1-3-5-7-9-11-13-15-17-19-21-23-25-27-29-28-26-24-22-20-18-16-14-
<b>InchiKey:</b>	UBMJSQAFNUWJEG-UHFFFAOYSA-N
<b>Formula:</b>	C29H52
<b>SMILES:</b>	C=CCCCCCCCCCCCCCCCCCCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	400.72
<b>CAS:</b>	18835-35-3

## Physical Properties

Property code	Value	Unit	Source
gf	281.14	kJ/mol	Joback Method
hf	-516.46	kJ/mol	Joback Method
hfus	69.59	kJ/mol	Joback Method
hvap	79.48	kJ/mol	Joback Method
log10ws	-11.82		Crippen Method
logp	11.335		Crippen Method
mcvol	415.170	ml/mol	McGowan Method
pc	637.69	kPa	Joback Method
rinpol	2894.00		NIST Webbook
rinpol	2877.00		NIST Webbook
rinpol	2888.00		NIST Webbook
rinpol	2884.70		NIST Webbook
tb	859.60	K	Joback Method
tc	1055.27	K	Joback Method
tf	414.83	K	Joback Method
vc	1.641	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1385.92	J/mol×K	859.60	Joback Method
cpg	1411.53	J/mol×K	892.21	Joback Method
cpg	1435.79	J/mol×K	924.82	Joback Method
cpg	1458.77	J/mol×K	957.43	Joback Method
cpg	1480.54	J/mol×K	990.05	Joback Method

cpg	1501.18	J/mol×K	1022.66	Joback Method
cpg	1520.75	J/mol×K	1055.27	Joback Method
dvisc	0.0012385	Paxs	414.83	Joback Method
dvisc	0.0004174	Paxs	488.96	Joback Method
dvisc	0.0001873	Paxs	563.09	Joback Method
dvisc	0.0001013	Paxs	637.22	Joback Method
dvisc	0.0000623	Paxs	711.34	Joback Method
dvisc	0.0000420	Paxs	785.47	Joback Method
dvisc	0.0000303	Paxs	859.60	Joback Method
hvapt	113.30	kJ/mol	584.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.16621e+01
Coeff. B	-3.16266e+03
Coeff. C	-2.61912e+02
Temperature range (K), min.	539.96
Temperature range (K), max.	759.92

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C18835353&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C18835353&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/74-635-5/1-Nonacosene.pdf>

Generated by Cheméo on 2024-04-26 15:16:09.536097793 +0000 UTC m=+16433818.456675105.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.