

# Isophthalic acid, 2-biphenyl nonyl ester

<b>Inchi:</b>	InChI=1S/C29H32O4/c1-2-3-4-5-6-7-13-21-32-28(30)24-17-14-18-25(22-24)29(31)33-27
<b>InchiKey:</b>	JZVWVYCGZVCITK-UHFFFAOYSA-N
<b>Formula:</b>	C29H32O4
<b>SMILES:</b>	CCCCCCCCCOC(=O)c1cccc(C(=O)Oc2ccccc2-c2ccccc2)c1
<b>Mol. weight [g/mol]:</b>	444.56

## Physical Properties

Property code	Value	Unit	Source
gf	43.43	kJ/mol	Joback Method
hf	-444.84	kJ/mol	Joback Method
hfus	57.78	kJ/mol	Joback Method
hvap	106.61	kJ/mol	Joback Method
log10ws	-9.75		Crippen Method
logp	7.480		Crippen Method
mcvol	363.070	ml/mol	McGowan Method
pc	1151.44	kPa	Joback Method
rinpol	3534.00		NIST Webbook
tb	1105.50	K	Joback Method
tc	1354.89	K	Joback Method
tf	665.21	K	Joback Method
vc	1.383	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1206.37	J/molxK	1105.50	Joback Method
cpg	1218.12	J/molxK	1147.06	Joback Method
cpg	1228.31	J/molxK	1188.63	Joback Method
cpg	1237.03	J/molxK	1230.19	Joback Method
cpg	1244.39	J/molxK	1271.76	Joback Method
cpg	1250.49	J/molxK	1313.32	Joback Method
cpg	1255.42	J/molxK	1354.89	Joback Method
dvisc	0.0001540	Paxs	665.21	Joback Method
dvisc	0.0000876	Paxs	738.59	Joback Method

dvisc	0.0000552	Paxs	811.97	Joback Method
dvisc	0.0000376	Paxs	885.36	Joback Method
dvisc	0.0000271	Paxs	958.74	Joback Method
dvisc	0.0000205	Paxs	1032.12	Joback Method
dvisc	0.0000161	Paxs	1105.50	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344564&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344564&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/74-650-8/Isophthalic-acid-2-biphenyl-nonyl-ester.pdf>

Generated by Cheméo on 2024-04-23 08:54:49.966149598 +0000 UTC m=+16151738.886726962.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.