

Succinic acid, 4-bromo-2,6-difluorobenzyl ethyl ester

Inchi:	InChI=1S/C13H13BrF2O4/c1-2-19-12(17)3-4-13(18)20-7-9-10(15)5-8(14)6-11(9)16/h5-6
InchiKey:	ZWDFVAXBVABPDC-UHFFFAOYSA-N
Formula:	C13H13BrF2O4
SMILES:	CCOC(=O)CCC(=O)OCc1c(F)cc(Br)cc1F
Mol. weight [g/mol]:	351.14

Physical Properties

Property code	Value	Unit	Source
gf	-701.04	kJ/mol	Joback Method
hf	-965.02	kJ/mol	Joback Method
hfus	39.32	kJ/mol	Joback Method
hvap	71.91	kJ/mol	Joback Method
log10ws	-4.41		Crippen Method
logp	3.114		Crippen Method
mcvol	206.190	ml/mol	McGowan Method
pc	2248.26	kPa	Joback Method
rinpol	1947.00		NIST Webbook
rinpol	1947.00		NIST Webbook
tb	755.74	K	Joback Method
tc	960.90	K	Joback Method
tf	505.55	K	Joback Method
vc	0.801	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	537.23	J/molxK	755.74	Joback Method
cpg	548.45	J/molxK	789.93	Joback Method
cpg	558.91	J/molxK	824.13	Joback Method
cpg	568.60	J/molxK	858.32	Joback Method
cpg	577.52	J/molxK	892.51	Joback Method
cpg	585.68	J/molxK	926.70	Joback Method
cpg	593.08	J/molxK	960.90	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U381151&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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