

# 7-Phenyl-1-heptyl chloride

<b>Other names:</b>	1-Chloro-7-phenylheptane
<b>Inchi:</b>	InChI=1S/C13H19Cl/c14-12-8-3-1-2-5-9-13-10-6-4-7-11-13/h4,6-7,10-11H,1-3,5,8-9,12H
<b>InchiKey:</b>	PNUBJDFTMKBZJK-UHFFFAOYSA-N
<b>Formula:</b>	C13H19Cl
<b>SMILES:</b>	C1CCCCCCCc1ccccc1
<b>Mol. weight [g/mol]:</b>	210.74
<b>CAS:</b>	71434-47-4

## Physical Properties

Property code	Value	Unit	Source
gf	159.06	kJ/mol	Joback Method
hf	-90.86	kJ/mol	Joback Method
hfus	27.66	kJ/mol	Joback Method
hvap	51.19	kJ/mol	Joback Method
log10ws	-4.52		Crippen Method
logp	4.418		Crippen Method
mcvol	182.510	ml/mol	McGowan Method
pc	2119.73	kPa	Joback Method
tb	560.95	K	Joback Method
tc	762.24	K	Joback Method
tf	292.61	K	Joback Method
vc	0.705	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	424.66	J/molxK	560.95	Joback Method
cpg	498.82	J/molxK	728.69	Joback Method
cpg	485.70	J/molxK	695.14	Joback Method
cpg	471.77	J/molxK	661.60	Joback Method
cpg	456.98	J/molxK	628.05	Joback Method
cpg	441.29	J/molxK	594.50	Joback Method
cpg	511.15	J/molxK	762.24	Joback Method
dvisc	0.0001924	Paxs	560.95	Joback Method

dvisc	0.0002517	Paxs	516.23	Joback Method
dvisc	0.0003465	Paxs	471.50	Joback Method
dvisc	0.0005100	Paxs	426.78	Joback Method
dvisc	0.0008218	Paxs	382.06	Joback Method
dvisc	0.0015028	Paxs	337.33	Joback Method
dvisc	0.0033048	Paxs	292.61	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C71434474&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C71434474&amp;Units=SI</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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