

# 3n-valeryl-4,5-dihydrophthalide

|                      |                                                                                   |
|----------------------|-----------------------------------------------------------------------------------|
| Inchi:               | InChI=1S/C13H18O/c1-2-3-4-9-13-12-8-6-5-7-11(12)10-14-13/h5,7,9H,2-4,6,8,10H2,1H3 |
| InchiKey:            | ZLAFQLXVGWHNRT-LCYFTJDESA-N                                                       |
| Formula:             | C13H18O                                                                           |
| SMILES:              | CCCCC=C1OCC2=C1CCC=C2                                                             |
| Mol. weight [g/mol]: | 190.28                                                                            |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | 159.20  | kJ/mol  | Joback Method  |
| hf            | -107.20 | kJ/mol  | Joback Method  |
| hfus          | 27.22   | kJ/mol  | Joback Method  |
| hvap          | 52.70   | kJ/mol  | Joback Method  |
| log10ws       | -4.20   |         | Crippen Method |
| logp          | 3.737   |         | Crippen Method |
| mcvol         | 165.280 | ml/mol  | McGowan Method |
| pc            | 2492.52 | kPa     | Joback Method  |
| rinpol        | 1767.00 |         | NIST Webbook   |
| rinpol        | 1767.00 |         | NIST Webbook   |
| tb            | 574.34  | K       | Joback Method  |
| tc            | 790.03  | K       | Joback Method  |
| tf            | 333.56  | K       | Joback Method  |
| vc            | 0.631   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 411.20    | J/mol×K | 574.34          | Joback Method |
| cpg           | 428.16    | J/mol×K | 610.29          | Joback Method |
| cpg           | 444.04    | J/mol×K | 646.24          | Joback Method |
| cpg           | 458.93    | J/mol×K | 682.18          | Joback Method |
| cpg           | 472.88    | J/mol×K | 718.13          | Joback Method |
| cpg           | 485.98    | J/mol×K | 754.08          | Joback Method |
| cpg           | 498.30    | J/mol×K | 790.03          | Joback Method |
| dvisc         | 0.0021356 | Pa×s    | 333.56          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0013346 | Pa×s | 373.69 | Joback Method |
| dvisc | 0.0009137 | Pa×s | 413.82 | Joback Method |
| dvisc | 0.0006689 | Pa×s | 453.95 | Joback Method |
| dvisc | 0.0005151 | Pa×s | 494.08 | Joback Method |
| dvisc | 0.0004125 | Pa×s | 534.21 | Joback Method |
| dvisc | 0.0003408 | Pa×s | 574.34 | Joback Method |

## Sources

|                        |                                                                                                                                           |
|------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R219324&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R219324&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                         |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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