

1-ISOPROPYL-3-TERT-BUTYLBENZENE

Inchi:	InChI=1S/C13H20/c1-10(2)11-7-6-8-12(9-11)13(3,4)5/h6-10H,1-5H3
InchiKey:	OWOLIQXOOMFSJE-UHFFFAOYSA-N
Formula:	C13H20
SMILES:	CC(C)c1cccc(C(C)(C)C)c1
Mol. weight [g/mol]:	176.30

Physical Properties

Property code	Value	Unit	Source
af	0.3805		Cheméo Relay (1.0)
gf	161.76	kJ/mol	Joback Method
hf	-101.68	kJ/mol	Cheméo Relay (1.0)
hfus	12.14	kJ/mol	Joback Method
hvap	58.34	kJ/mol	Cheméo Relay (1.0)
ie	8.32	eV	Cheméo Relay (1.0)
log10ws	-5.31		Cheméo Relay (1.0)
logp	4.107		Crippen Method
mcvol	170.270	ml/mol	McGowan Method
pc	2218.71	kPa	Joback Method
ripol	1397.00		NIST Webbook
ripol	1404.00		NIST Webbook
ripol	1431.00		NIST Webbook
ripol	1414.00		NIST Webbook
ripol	1406.00		NIST Webbook
tb	482.70 ± 1.00	K	NIST Webbook
tc	698.74	K	Cheméo Relay (1.0)
tf	238.34	K	Cheméo Relay (1.0)
vc	0.632	m3/kmol	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	396.59	J/mol×K	524.83	Joback Method
cpg	493.91	J/mol×K	739.67	Joback Method
cpg	480.24	J/mol×K	703.86	Joback Method

cpg	465.63	J/mol×K	668.05	Joback Method
cpg	450.02	J/mol×K	632.25	Joback Method
cpg	433.35	J/mol×K	596.44	Joback Method
cpg	415.56	J/mol×K	560.64	Joback Method
dvisc	0.0005231	Pa×s	393.73	Joback Method
dvisc	0.0001675	Pa×s	524.83	Joback Method
dvisc	0.0002285	Pa×s	481.13	Joback Method
dvisc	0.0003317	Pa×s	437.43	Joback Method
dvisc	0.0050928	Pa×s	262.63	Joback Method
dvisc	0.0009242	Pa×s	350.03	Joback Method
dvisc	0.0019209	Pa×s	306.33	Joback Method

Sources

Cheméo Relay (1.0):

Cheméo Relay (1.0, beta):

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C20033129&Units=SI>

Joback Method:

https://en.wikipedia.org/wiki/Joback_method

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Legend

af:	Acentric Factor
cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature

tc: Critical Temperature
tf: Normal melting (fusion) point
vc: Critical Volume

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