

# Benzene, 1,4-dimethyl-2-pentyl

Inchi:	InChI=1S/C13H20/c1-4-5-6-7-13-10-11(2)8-9-12(13)3/h8-10H,4-7H2,1-3H3
InchiKey:	ROQGPVNDWWQAW-UHFFFAOYSA-N
Formula:	C13H20
SMILES:	CCCCC1cc(C)ccc1C
Mol. weight [g/mol]:	176.30

## Physical Properties

Property code	Value	Unit	Source
gf	151.73	kJ/mol	Joback Method
hf	-98.06	kJ/mol	Joback Method
hfus	22.69	kJ/mol	Joback Method
hvap	48.13	kJ/mol	Joback Method
log10ws	-4.48		Crippen Method
logp	4.036		Crippen Method
mcvol	170.270	ml/mol	McGowan Method
pc	2133.46	kPa	Joback Method
ripol	1609.00		NIST Webbook
ripol	1648.00		NIST Webbook
ripol	1608.00		NIST Webbook
ripol	1594.00		NIST Webbook
ripol	1624.00		NIST Webbook
ripol	1632.00		NIST Webbook
tb	533.48	K	Joback Method
tc	732.26	K	Joback Method
tf	287.73	K	Joback Method
vc	0.655	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	393.98	J/molxK	533.48	Joback Method
cpg	410.89	J/molxK	566.61	Joback Method
cpg	426.98	J/molxK	599.74	Joback Method
cpg	442.26	J/molxK	632.87	Joback Method

cpg	456.76	J/molxK	666.00	Joback Method
cpg	470.52	J/molxK	699.13	Joback Method
cpg	483.54	J/molxK	732.26	Joback Method
dvisc	0.0019117	Paxs	287.73	Joback Method
dvisc	0.0010134	Paxs	328.69	Joback Method
dvisc	0.0006184	Paxs	369.65	Joback Method
dvisc	0.0004164	Paxs	410.61	Joback Method
dvisc	0.0003012	Paxs	451.56	Joback Method
dvisc	0.0002300	Paxs	492.52	Joback Method
dvisc	0.0001830	Paxs	533.48	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R550066&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R550066&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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