

# «alpha»-Butylcinnamic aldehyde

<b>Other names:</b>	«alpha»-butylcinnamaldehyde
<b>Inchi:</b>	InChI=1S/C13H16O/c1-2-3-7-13(11-14)10-12-8-5-4-6-9-12/h4-6,8-11H,2-3,7H2,1H3/b13
<b>InchiKey:</b>	GFBCBQNBXRPRQD-JLHYYAGUSA-N
<b>Formula:</b>	C13H16O
<b>SMILES:</b>	CCCCC(C=O)=Cc1cccc1
<b>Mol. weight [g/mol]:</b>	188.27
<b>CAS:</b>	7492-44-6

## Physical Properties

Property code	Value	Unit	Source
gf	143.14	kJ/mol	Joback Method
hf	-53.27	kJ/mol	Joback Method
hfus	24.65	kJ/mol	Joback Method
hvap	53.57	kJ/mol	Joback Method
log10ws	-3.67		Crippen Method
logp	3.459		Crippen Method
mcvol	167.540	ml/mol	McGowan Method
pc	2490.03	kPa	Joback Method
rinpol	1535.00		NIST Webbook
rinpol	1535.00		NIST Webbook
ripol	2160.00		NIST Webbook
ripol	2160.00		NIST Webbook
tb	576.22	K	Joback Method
tc	788.88	K	Joback Method
tf	285.65	K	Joback Method
vc	0.653	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	397.65	J/molxK	576.22	Joback Method
cpg	413.25	J/molxK	611.66	Joback Method
cpg	427.86	J/molxK	647.11	Joback Method
cpg	441.53	J/molxK	682.55	Joback Method

cpg	454.31	J/mol×K	717.99	Joback Method
cpg	466.27	J/mol×K	753.44	Joback Method
cpg	477.46	J/mol×K	788.88	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7492446&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7492446&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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