

Carbonic acid, but-3-en-1-yl hexyl ester

| | |
|-----------------------------|--|
| Inchi: | InChI=1S/C11H20O3/c1-3-5-7-8-10-14-11(12)13-9-6-4-2/h4H,2-3,5-10H2,1H3 |
| InchiKey: | DEHFDEBDQQOIHU-UHFFFAOYSA-N |
| Formula: | C11H20O3 |
| SMILES: | C=CCCOC(=O)OCCCCC |
| Mol. weight [g/mol]: | 200.27 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|---------|----------------|
| gf | -209.34 | kJ/mol | Joback Method |
| hf | -521.96 | kJ/mol | Joback Method |
| hfus | 26.94 | kJ/mol | Joback Method |
| hvap | 50.98 | kJ/mol | Joback Method |
| log10ws | -3.21 | | Crippen Method |
| logp | 3.296 | | Crippen Method |
| mcvol | 174.860 | ml/mol | McGowan Method |
| pc | 2060.49 | kPa | Joback Method |
| rinpol | 1338.00 | | NIST Webbook |
| rinpol | 1338.00 | | NIST Webbook |
| tb | 546.47 | K | Joback Method |
| tc | 719.93 | K | Joback Method |
| tf | 306.36 | K | Joback Method |
| vc | 0.674 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 422.56 | J/molxK | 546.47 | Joback Method |
| cpg | 436.77 | J/molxK | 575.38 | Joback Method |
| cpg | 450.44 | J/molxK | 604.29 | Joback Method |
| cpg | 463.57 | J/molxK | 633.20 | Joback Method |
| cpg | 476.16 | J/molxK | 662.11 | Joback Method |
| cpg | 488.21 | J/molxK | 691.02 | Joback Method |
| cpg | 499.74 | J/molxK | 719.93 | Joback Method |
| dvisc | 0.0021653 | Paxs | 306.36 | Joback Method |

| | | | | |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0011102 | Paxs | 346.38 | Joback Method |
| dvisc | 0.0006537 | Paxs | 386.40 | Joback Method |
| dvisc | 0.0004251 | Paxs | 426.42 | Joback Method |
| dvisc | 0.0002977 | Paxs | 466.43 | Joback Method |
| dvisc | 0.0002205 | Paxs | 506.45 | Joback Method |
| dvisc | 0.0001707 | Paxs | 546.47 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U383227&Units=SI |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

Latest version available from:

<https://www.chemeo.com/cid/74-931-6/Carbonic-acid-but-3-en-1-yl-hexyl-ester.pdf>

Generated by Cheméo on 2025-12-05 15:14:45.913858424 +0000 UTC m=+4695883.443899078.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.