

# Isobicyclogermacral

**Inchi:** InChI=1S/C15H22O/c1-11-5-4-6-12(10-16)9-14-13(8-7-11)15(14,2)3/h5,9-10,13-14H,4,6  
**InchiKey:** BLCUVJCHWZPQCX-IOGNAORRSA-N  
**Formula:** C15H22O  
**SMILES:** CC1=CCCC(G=O)=CC2C(CC1)C2(C)C  
**Mol. weight [g/mol]:** 218.33

## Physical Properties

Property code	Value	Unit	Source
gf	64.36	kJ/mol	Joback Method
hf	-236.19	kJ/mol	Joback Method
hfus	19.10	kJ/mol	Joback Method
hvap	56.84	kJ/mol	Joback Method
log10ws	-4.15		Crippen Method
logp	3.904		Crippen Method
mcvol	193.460	ml/mol	McGowan Method
pc	2173.43	kPa	Joback Method
ripol	2325.00		NIST Webbook
ripol	2325.00		NIST Webbook
ripol	2325.00		NIST Webbook
ripol	2325.00		NIST Webbook
ripol	2325.00		NIST Webbook
ripol	2325.00		NIST Webbook
ripol	2325.00		NIST Webbook
ripol	2325.00		NIST Webbook
tb	629.94	K	Joback Method
tc	858.86	K	Joback Method
tf	365.31	K	Joback Method
vc	0.736	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	530.57	J/molxK	629.94	Joback Method
cpg	551.52	J/molxK	668.09	Joback Method

cpg	571.23	J/mol×K	706.25	Joback Method
cpg	589.86	J/mol×K	744.40	Joback Method
cpg	607.55	J/mol×K	782.55	Joback Method
cpg	624.44	J/mol×K	820.71	Joback Method
cpg	640.69	J/mol×K	858.86	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R323334&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R323334&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/74-945-1/Isobicyclogermacral.pdf>

Generated by Cheméo on 2024-04-25 05:24:52.655488176 +0000 UTC m=+16311941.576065488.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.