

# «beta»-Vetispirene

**Other names:**

1,8a-dimethyl-7-propan-2-ylidene-1,2,6,8-tetrahydronaphthalene  
Naphthalene, 1,2,6,7,8,8a-hexahydro-1,8a-dimethyl-7-(1-methylethylidene)-,  
(1R,8aS)-4«beta»H,5«alpha»-Eremophila-1,7(11),9-triene  
«beta»-Vatirenene  
«beta»-Vetivenene

**Inchi:**

InChI=1S/C15H22/c1-11(2)13-8-9-14-7-5-6-12(3)15(14,4)10-13/h5,7,9,12H,6,8,10H2,1-4

**InchiKey:**

QSUQBKPPUWLTH-DOMZBBRYSA-N

**Formula:**

C15H22

**SMILES:**

CC(C)=C1CC=C2C=CCC(C)C2(C)C1

**Mol. weight [g/mol]:**

202.34

## Physical Properties

Property code	Value	Unit	Source
gf	230.23	kJ/mol	Joback Method
hf	-46.40	kJ/mol	Joback Method
hfus	17.25	kJ/mol	Joback Method
hvap	50.46	kJ/mol	Joback Method
log10ws	-4.97		Crippen Method
logp	4.645		Crippen Method
mcvol	187.590	ml/mol	McGowan Method
pc	2139.38	kPa	Joback Method
rinpol	1563.00		NIST Webbook
rinpol	1488.00		NIST Webbook
rinpol	1488.00		NIST Webbook
rinpol	1554.00		NIST Webbook
rinpol	1547.00		NIST Webbook
rinpol	1554.00		NIST Webbook
rinpol	1542.00		NIST Webbook
rinpol	1527.00		NIST Webbook
rinpol	1540.00		NIST Webbook
rinpol	1507.00		NIST Webbook
rinpol	1515.00		NIST Webbook
rinpol	1544.00		NIST Webbook
rinpol	1544.00		NIST Webbook
rinpol	1574.00		NIST Webbook
ripol	1885.00		NIST Webbook
ripol	1852.00		NIST Webbook

ripol	1864.00		NIST Webbook
ripol	1868.00		NIST Webbook
ripol	1885.00		NIST Webbook
ripol	1868.00		NIST Webbook
tb	583.22	K	Joback Method
tc	812.62	K	Joback Method
tf	314.95	K	Joback Method
vc	0.712	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	480.83	J/mol×K	583.22	Joback Method
cpg	502.10	J/mol×K	621.45	Joback Method
cpg	522.00	J/mol×K	659.69	Joback Method
cpg	540.71	J/mol×K	697.92	Joback Method
cpg	558.40	J/mol×K	736.15	Joback Method
cpg	575.26	J/mol×K	774.38	Joback Method
cpg	591.45	J/mol×K	812.62	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C27840400&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C27840400&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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