

# Drima-7,9(11)-diene

<b>Other names:</b>	DRIMA-7,9-DIENE
<b>Inchi:</b>	InChI=1S/C15H24/c1-11-7-8-13-14(3,4)9-6-10-15(13,5)12(11)2/h7,13H,2,6,8-10H2,1,3-5
<b>InchiKey:</b>	IQLMHLRIRXSXLQ-UHFFFAOYSA-N
<b>Formula:</b>	C15H24
<b>SMILES:</b>	<chem>C=C1C(C)=CCC2C(C)(C)CCCC12C</chem>
<b>Mol. weight [g/mol]:</b>	204.35

## Physical Properties

Property code	Value	Unit	Source
gf	203.24	kJ/mol	Joback Method
hf	-91.28	kJ/mol	Joback Method
hfus	10.62	kJ/mol	Joback Method
hvap	48.00	kJ/mol	Joback Method
log10ws	-4.87		Crippen Method
logp	4.725		Crippen Method
mcvol	191.890	ml/mol	McGowan Method
pc	2064.24	kPa	Joback Method
rinpol	1469.00		NIST Webbook
rinpol	1471.00		NIST Webbook
rinpol	1473.00		NIST Webbook
rinpol	1473.00		NIST Webbook
rinpol	1473.00		NIST Webbook
rinpol	1473.00		NIST Webbook
rinpol	1456.00		NIST Webbook
rinpol	1476.00		NIST Webbook
rinpol	1470.00		NIST Webbook
rinpol	1469.00		NIST Webbook
rinpol	1470.00		NIST Webbook
rinpol	1469.00		NIST Webbook
rinpol	1469.00		NIST Webbook
rinpol	1468.00		NIST Webbook
rinpol	1461.00		NIST Webbook
rinpol	1469.00		NIST Webbook
rinpol	1473.00		NIST Webbook
rinpol	1461.00		NIST Webbook
rinpol	1473.00		NIST Webbook
rinpol	1488.00		NIST Webbook

ripol	1469.00		NIST Webbook
ripol	1677.00		NIST Webbook
ripol	1677.00		NIST Webbook
ripol	1677.00		NIST Webbook
ripol	1668.00		NIST Webbook
ripol	1668.00		NIST Webbook
tb	572.27	K	Joback Method
tc	799.19	K	Joback Method
tf	351.13	K	Joback Method
vc	0.723	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	498.67	J/mol×K	572.27	Joback Method
cpg	520.98	J/mol×K	610.09	Joback Method
cpg	541.90	J/mol×K	647.91	Joback Method
cpg	561.68	J/mol×K	685.73	Joback Method
cpg	580.56	J/mol×K	723.55	Joback Method
cpg	598.80	J/mol×K	761.37	Joback Method
cpg	616.65	J/mol×K	799.19	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R234492&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R234492&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>r<sub>ipol</sub>:</b>	Polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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