

# Benzeneethanamine, N-(phenylmethylene)-

<b>Other names:</b>	Phenethylamine, N-benzylidene- N-Benzylidene-«beta»-phenethylamine N-Benzylidene-«beta»-phenylethylamine N-Benzylidene phenylethylamine
<b>Inchi:</b>	InChI=1S/C15H15N/c1-3-7-14(8-4-1)11-12-16-13-15-9-5-2-6-10-15/h1-10,13H,11-12H2
<b>InchiKey:</b>	OObAKFDIGIHHOM-UHFFFAOYSA-N
<b>Formula:</b>	C15H15N
<b>SMILES:</b>	<chem>C(=NCCc1ccccc1)c1ccccc1</chem>
<b>Mol. weight [g/mol]:</b>	209.29
<b>CAS:</b>	3240-95-7

## Physical Properties

Property code	Value	Unit	Source
hf	202.35	kJ/mol	Joback Method
hvap	56.85	kJ/mol	Joback Method
log10ws	-3.57		Crippen Method
logp	3.348		Crippen Method
mcvol	180.370	ml/mol	McGowan Method
pc	2265.42	kPa	Joback Method
ripol	2560.00		NIST Webbook
tb	672.64	K	Joback Method
tc	922.22	K	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3240957&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3240957&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

# Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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