

# Acridine, 9,10-dihydro-9,9-dimethyl-

<b>Other names:</b>	Acridan, 9,9-dimethyl- 9,9-Dimethyl-9,10-dihydroacridine 9,9-Dimethylacridan 9,10-Dihydro-9,9-dimethylacridine
<b>Inchi:</b>	InChI=1S/C15H15N/c1-15(2)11-7-3-5-9-13(11)16-14-10-6-4-8-12(14)15/h3-10,16H,1-2H3
<b>InchiKey:</b>	JSEQNGYLWKBMI-UHFFFAOYSA-N
<b>Formula:</b>	C15H15N
<b>SMILES:</b>	CC1(C)c2ccccc2Nc2ccccc21
<b>Mol. weight [g/mol]:</b>	209.29
<b>CAS:</b>	6267-02-3

## Physical Properties

Property code	Value	Unit	Source
gf	436.05	kJ/mol	Joback Method
hf	229.20	kJ/mol	Joback Method
hfus	25.44	kJ/mol	Joback Method
hvap	60.21	kJ/mol	Joback Method
log10ws	-4.13		Crippen Method
logp	4.069		Crippen Method
mcvol	173.810	ml/mol	McGowan Method
pc	2918.68	kPa	Joback Method
rinpol	1906.00		NIST Webbook
tb	657.18	K	Joback Method
tc	915.26	K	Joback Method
tf	487.08	K	Joback Method
vc	0.659	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	447.01	J/molxK	657.18	Joback Method
cpg	463.72	J/molxK	700.19	Joback Method
cpg	479.41	J/molxK	743.21	Joback Method
cpg	494.33	J/molxK	786.22	Joback Method

cpg	508.77	J/mol×K	829.24	Joback Method
cpg	523.00	J/mol×K	872.25	Joback Method
cpg	537.27	J/mol×K	915.26	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6267023&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6267023&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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