

# 2(3H)-Phenanthrenone, 4,4a,9,10-tetrahydro-4a-methyl-

Other names:

2(3H)-Phenanthrone, 4,4a,9,10-tetrahydro-4a-methyl-  
4a-Methyl-4,4a,9,10-tetrahydro-2(3H)-phenanthrenone  
4,4a,9,10-Tetrahydro-4a-methyl-2(3H)-phenanthrenone

Inchi:

InChI=1S/C15H16O/c1-15-9-8-13(16)10-12(15)7-6-11-4-2-3-5-14(11)15/h2-5,10H,6-9H2

InchiKey:

FSPGJSRKHCJARC-UHFFFAOYSA-N

Formula:

C15H16O

SMILES:

CC12CCC(=O)C=C1CCc1cccc12

Mol. weight [g/mol]:

212.29

CAS:

6606-34-4

## Physical Properties

Property code	Value	Unit	Source
gf	175.46	kJ/mol	Joback Method
hf	-50.40	kJ/mol	Joback Method
hfus	13.30	kJ/mol	Joback Method
hvap	56.45	kJ/mol	Joback Method
log10ws	-3.77		Crippen Method
logp	3.180		Crippen Method
mcvol	174.000	ml/mol	McGowan Method
pc	2775.92	kPa	Joback Method
rinsol	288.30		NIST Webbook
tb	673.15	K	Joback Method
tc	935.55	K	Joback Method
tf	436.23	K	Joback Method
vc	0.657	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	475.96	J/mol×K	673.15	Joback Method
cpg	494.74	J/mol×K	716.88	Joback Method
cpg	512.40	J/mol×K	760.62	Joback Method
cpg	529.17	J/mol×K	804.35	Joback Method
cpg	545.33	J/mol×K	848.08	Joback Method

cpg	561.10	J/mol×K	891.82	Joback Method
cpg	576.75	J/mol×K	935.55	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6606344&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6606344&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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