

# Benzeneethanol, «beta»-methyl-«beta»-phenyl-

<b>Other names:</b>	1-Propanol, 2,2-diphenyl- 2,2-diphenylpropan-1-ol
<b>Inchi:</b>	InChI=1S/C15H16O/c1-15(12-16,13-8-4-2-5-9-13)14-10-6-3-7-11-14/h2-11,16H,12H2,1H
<b>InchiKey:</b>	WAUNMVYXQAKNLE-UHFFFAOYSA-N
<b>Formula:</b>	C15H16O
<b>SMILES:</b>	CC(CO)(c1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	212.29
<b>CAS:</b>	74421-26-4

## Physical Properties

Property code	Value	Unit	Source
gf	166.26	kJ/mol	Joback Method
hf	-40.85	kJ/mol	Joback Method
hfus	19.36	kJ/mol	Joback Method
h vap	68.92	kJ/mol	Joback Method
log10ws	-3.35		Crippen Method
logp	2.985		Crippen Method
m cvol	180.560	ml/mol	McGowan Method
pc	2814.34	kPa	Joback Method
tb	684.91	K	Joback Method
tc	912.98	K	Joback Method
tf	374.89	K	Joback Method
vc	0.667	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	482.88	J/mol×K	684.91	Joback Method
cpg	546.08	J/mol×K	874.97	Joback Method
cpg	535.40	J/mol×K	836.96	Joback Method
cpg	523.83	J/mol×K	798.95	Joback Method
cpg	511.29	J/mol×K	760.93	Joback Method
cpg	497.67	J/mol×K	722.92	Joback Method
cpg	555.99	J/mol×K	912.98	Joback Method

dvisc	0.0000360	Paxs	684.91	Joback Method
dvisc	0.0000562	Paxs	633.24	Joback Method
dvisc	0.0000951	Paxs	581.57	Joback Method
dvisc	0.0001784	Paxs	529.90	Joback Method
dvisc	0.0003832	Paxs	478.23	Joback Method
dvisc	0.0009906	Paxs	426.56	Joback Method
dvisc	0.0033272	Paxs	374.89	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C74421264&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C74421264&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/75-206-0/Benzeneethanol-beta-methyl-beta-phenyl.pdf>

Generated by Cheméo on 2024-04-24 03:20:31.674008805 +0000 UTC m=+16218080.594586117.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.