

# 1-Adamantanemethylamine, «alpha»-methyl-

<b>Other names:</b>	(dl)-rimantadine .alpha.-methyltricyclo[3.3.1.1(3,7)]decane-1-methanamine 1-(adamantan-1-yl)ethan-1-amine Rimantadine Tricyclo(3.3.1.1 «alpha»-Methyl-1-adamantanemethylamine «alpha»-Methyladamantanemethylamine
<b>Inchi:</b>	InChI=1S/C12H21N/c1-8(13)12-5-9-2-10(6-12)4-11(3-9)7-12/h8-11H,2-7,13H2,1H3
<b>InchiKey:</b>	UBCHPRBFMUDMNC-UHFFFAOYSA-N
<b>Formula:</b>	C12H21N
<b>SMILES:</b>	CC(N)C12CC3CC(CC(C3)C1)C2
<b>Mol. weight [g/mol]:</b>	179.30
<b>CAS:</b>	13392-28-4

## Physical Properties

Property code	Value	Unit	Source
gf	271.12	kJ/mol	Joback Method
hf	-55.36	kJ/mol	Joback Method
hfus	15.59	kJ/mol	Joback Method
hvap	51.01	kJ/mol	Joback Method
log10ws	-3.11		Crippen Method
logp	2.550		Crippen Method
mcvol	157.340	ml/mol	McGowan Method
pc	2835.36	kPa	Joback Method
tb	566.11	K	Joback Method
tc	797.40	K	Joback Method
tf	363.22	K	Joback Method
vc	0.591	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	440.95	J/molxK	566.11	Joback Method
cpg	462.18	J/molxK	604.66	Joback Method

cpg	481.81	J/mol×K	643.21	Joback Method
cpg	500.10	J/mol×K	681.76	Joback Method
cpg	517.27	J/mol×K	720.30	Joback Method
cpg	533.55	J/mol×K	758.85	Joback Method
cpg	549.18	J/mol×K	797.40	Joback Method
pvap	3.00e-03	kPa	298.15	The Vaporization Enthalpies and Vapor Pressures of Some Primary Amines of Pharmaceutical Importance by Correlation Gas Chromatography

## Sources

The Vaporization Enthalpies and Vapor Pressures of Some Primary Amines of Pharmaceutical Importance by Correlation Gas Chromatography: McGowan Method:

<https://www.doi.org/10.1021/je400498a>

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

NIST Webbook:

<http://link.springer.com/article/10.1007/BF02311772>

Crippen Method:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C13392284&Units=SI>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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