

1,3-Dioxane, 2-ethyl-2methyl-4-pentyl, 2R,4R

Inchi:

InChI=1S/C12H24O2/c1-4-6-7-8-11-9-10-13-12(3,5-2)14-11/h11H,4-10H2,1-3H3/t11-,12-

InchiKey:

DURNOWZZDBQNIF-RYUDHWPBXSA-N

Formula:

C12H24O2

SMILES:

CCCCC1CCOC(C)(CC)O1

Mol. weight [g/mol]:

200.32

Physical Properties

Property code	Value	Unit	Source
gf	-110.83	kJ/mol	Joback Method
hf	-505.79	kJ/mol	Joback Method
hfus	29.40	kJ/mol	Joback Method
hvap	50.30	kJ/mol	Joback Method
log10ws	-3.63		Crippen Method
logp	3.498		Crippen Method
mcvol	180.820	ml/mol	McGowan Method
pc	2121.68	kPa	Joback Method
ripol	1516.00		NIST Webbook
tb	542.98	K	Joback Method
tc	740.89	K	Joback Method
tf	305.18	K	Joback Method
vc	0.679	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	466.39	J/mol×K	542.98	Joback Method
cpg	486.11	J/mol×K	575.96	Joback Method
cpg	504.78	J/mol×K	608.95	Joback Method
cpg	522.50	J/mol×K	641.93	Joback Method
cpg	539.34	J/mol×K	674.92	Joback Method
cpg	555.39	J/mol×K	707.90	Joback Method
cpg	570.74	J/mol×K	740.89	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R191805&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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