

4-thiatridecane

Other names:	nonyl propyl sulfide
Inchi:	InChI=1S/C12H26S/c1-3-5-6-7-8-9-10-12-13-11-4-2/h3-12H2,1-2H3
InchiKey:	AQAOPNMAEGBYHI-UHFFFAOYSA-N
Formula:	C12H26S
SMILES:	CCCCCCCCCSCCC
Mol. weight [g/mol]:	202.40
CAS:	62103-66-6

Physical Properties

Property code	Value	Unit	Source
af	0.5684		Cheméo Relay (1.0)
gf	83.28	kJ/mol	Joback Method
hf	-260.98	kJ/mol	Cheméo Relay (1.0)
hfus	30.97	kJ/mol	Joback Method
hvap	70.69	kJ/mol	Cheméo Relay (1.0)
ie	8.39	eV	Cheméo Relay (1.0)
log10ws	-5.87		Cheméo Relay (1.0)
logp	4.880		Crippen Method
mcvol	196.290	ml/mol	McGowan Method
pc	1806.16	kPa	Joback Method
tb	527.15	K	Cheméo Relay (1.0)
tc	713.24	K	Cheméo Relay (1.0)
tf	250.57	K	Cheméo Relay (1.0)
vc	0.759	m3/kmol	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	472.53	J/mol×K	542.74	Joback Method
cpg	489.74	J/mol×K	572.48	Joback Method
cpg	506.22	J/mol×K	602.21	Joback Method
cpg	521.99	J/mol×K	631.95	Joback Method
cpg	537.07	J/mol×K	661.69	Joback Method
cpg	551.47	J/mol×K	691.42	Joback Method

cpg

565.21

J/mol×K

721.16

Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.60078e+01
Coeff. B	-4.94017e+03
Coeff. C	-8.99520e+01
Temperature range (K), min.	404.21
Temperature range (K), max.	551.81

Sources

NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=C62103666&Units=SI>

Joback Method: https://en.wikipedia.org/wiki/Joback_method

McGowan Method: <http://link.springer.com/article/10.1007/BF02311772>

Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws

Cheméo Relay (1.0):

Cheméo Relay (1.0, beta):

The Yaws Handbook of Vapor Pressure: <https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

Legend

af: Acentric Factor
cpg: Ideal gas heat capacity
gf: Standard Gibbs free energy of formation
hf: Enthalpy of formation at standard conditions
hfus: Enthalpy of fusion at standard conditions
hvap: Enthalpy of vaporization at standard conditions
ie: Ionization energy
log10ws: Log10 of Water solubility in mol/l
logp: Octanol/Water partition coefficient

mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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