

# Naphthalene, 5-ethyl-1,2,3,4-tetrahydro-

<b>Other names:</b>	5-Ethyltetralin 5-Ethyltetraline Tetraline, 5-ethyl
<b>Inchi:</b>	InChI=1S/C12H16/c1-2-10-7-5-8-11-6-3-4-9-12(10)11/h5,7-8H,2-4,6,9H2,1H3
<b>InchiKey:</b>	WHCKFZOMBDMRLO-UHFFFAOYSA-N
<b>Formula:</b>	C12H16
<b>SMILES:</b>	CCc1cccc2c1CCCC2
<b>Mol. weight [g/mol]:</b>	160.26
<b>CAS:</b>	42775-75-7

## Physical Properties

Property code	Value	Unit	Source
chl	-6569.00	kJ/mol	NIST Webbook
gf	199.67	kJ/mol	Joback Method
hf	9.56	kJ/mol	Joback Method
hfus	15.06	kJ/mol	Joback Method
hvap	46.30	kJ/mol	Joback Method
log10ws	-3.77		Crippen Method
logp	3.128		Crippen Method
mcvol	145.320	ml/mol	McGowan Method
pc	2826.33	kPa	Joback Method
rinpol	1362.00		NIST Webbook
rinpol	1354.00		NIST Webbook
rinpol	1354.00		NIST Webbook
tb	521.17 ± 0.30	K	NIST Webbook
tc	752.00	K	Joback Method
tf	228.60 ± 0.20	K	NIST Webbook
vc	0.549	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	332.17	J/molxK	526.28	Joback Method
cpg	349.96	J/molxK	563.90	Joback Method

cpg	366.60	J/molxK	601.52	Joback Method
cpg	382.14	J/molxK	639.14	Joback Method
cpg	396.66	J/molxK	676.76	Joback Method
cpg	410.22	J/molxK	714.38	Joback Method
cpg	422.89	J/molxK	752.00	Joback Method
dvisc	0.0021455	Paxs	295.12	Joback Method
dvisc	0.0012818	Paxs	333.65	Joback Method
dvisc	0.0008520	Paxs	372.17	Joback Method
dvisc	0.0006114	Paxs	410.70	Joback Method
dvisc	0.0004645	Paxs	449.23	Joback Method
dvisc	0.0003685	Paxs	487.75	Joback Method
dvisc	0.0003024	Paxs	526.28	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C42775757&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C42775757&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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