

# Tricyclo[5.2.1.0(2.6)]dec-3-ene, 3-methyl-9-methylene

Inchi:	InChI=1S/C12H16/c1-7-3-4-10-9-5-8(2)11(6-9)12(7)10/h3,9-12H,2,4-6H2,1H3
InchiKey:	PPYRKCDWMMCCSX-UHFFFAOYSA-N
Formula:	C12H16
SMILES:	C=C1CC2CC1C1C(C)=CCC21
Mol. weight [g/mol]:	160.26

## Physical Properties

Property code	Value	Unit	Source
gf	286.01	kJ/mol	Joback Method
hf	31.44	kJ/mol	Joback Method
hfus	19.89	kJ/mol	Joback Method
hvap	43.02	kJ/mol	Joback Method
log10ws	-3.27		Crippen Method
logp	3.165		Crippen Method
mcvol	138.760	ml/mol	McGowan Method
pc	2635.25	kPa	Joback Method
rinpol	1161.60		NIST Webbook
rinpol	1206.30		NIST Webbook
rinpol	1176.20		NIST Webbook
rinpol	1186.30		NIST Webbook
rinpol	1176.20		NIST Webbook
rinpol	1169.60		NIST Webbook
tb	497.08	K	Joback Method
tc	709.93	K	Joback Method
tf	298.02	K	Joback Method
vc	0.539	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	334.82	J/mol×K	497.08	Joback Method
cpg	418.39	J/mol×K	674.45	Joback Method
cpg	403.86	J/mol×K	638.98	Joback Method
cpg	388.32	J/mol×K	603.50	Joback Method

cpg	371.69	J/mol×K	568.03	Joback Method
cpg	353.89	J/mol×K	532.55	Joback Method
cpg	432.00	J/mol×K	709.93	Joback Method
dvisc	0.0014471	Paxs	497.08	Joback Method
dvisc	0.0013581	Paxs	463.90	Joback Method
dvisc	0.0012622	Paxs	430.73	Joback Method
dvisc	0.0011589	Paxs	397.55	Joback Method
dvisc	0.0010476	Paxs	364.37	Joback Method
dvisc	0.0009280	Paxs	331.20	Joback Method
dvisc	0.0008002	Paxs	298.02	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R298065&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R298065&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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