

# 5',6',7',8'-Tetrahydro-2'-acetonaphthone

<b>Other names:</b>	6-Acetyl tetralin Ethanone, 1-(5,6,7,8-tetrahydro-2-naphthalenyl)- 1-(5,6,7,8-tetrahydro-2-naphthyl)ethan-1-one
<b>Inchi:</b>	InChI=1S/C12H14O/c1-9(13)11-7-6-10-4-2-3-5-12(10)8-11/h6-8H,2-5H2,1H3
<b>InchiKey:</b>	VEPUKHYYQNXSSKV-UHFFFAOYSA-N
<b>Formula:</b>	C12H14O
<b>SMILES:</b>	CC(=O)c1ccc2c(c1)CCCC2
<b>Mol. weight [g/mol]:</b>	174.24
<b>CAS:</b>	774-55-0

## Physical Properties

Property code	Value	Unit	Source
gf	70.75	kJ/mol	Joback Method
hf	-103.02	kJ/mol	Joback Method
hfus	16.66	kJ/mol	Joback Method
hvap	53.05	kJ/mol	Joback Method
log10ws	-3.63		Crippen Method
logp	2.768		Crippen Method
mcvol	146.890	ml/mol	McGowan Method
pc	3025.61	kPa	Joback Method
tb	580.15	K	Joback Method
tc	814.20	K	Joback Method
tf	345.05	K	Joback Method
vc	0.555	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	352.93	J/molxK	580.15	Joback Method
cpg	369.06	J/molxK	619.16	Joback Method
cpg	384.06	J/molxK	658.17	Joback Method
cpg	398.00	J/molxK	697.17	Joback Method
cpg	410.94	J/molxK	736.18	Joback Method
cpg	422.96	J/molxK	775.19	Joback Method

cpg	434.12	J/mol×K	814.20	Joback Method
dvisc	0.0021469	Paxs	345.05	Joback Method
dvisc	0.0013528	Paxs	384.23	Joback Method
dvisc	0.0009285	Paxs	423.42	Joback Method
dvisc	0.0006792	Paxs	462.60	Joback Method
dvisc	0.0005218	Paxs	501.78	Joback Method
dvisc	0.0004164	Paxs	540.97	Joback Method
dvisc	0.0003426	Paxs	580.15	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C774550&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C774550&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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