

# 4,7-Dimethyl-1-tetralone

<b>Other names:</b>	4,7-Dimethyltetral-1-one
<b>Inchi:</b>	InChI=1S/C12H14O/c1-8-3-5-10-9(2)4-6-12(13)11(10)7-8/h3,5,7,9H,4,6H2,1-2H3
<b>InchiKey:</b>	SQESYXTWWGWCFK-UHFFFAOYSA-N
<b>Formula:</b>	C12H14O
<b>SMILES:</b>	<chem>Cc1ccc2c(c1)C(=O)CCC2C</chem>
<b>Mol. weight [g/mol]:</b>	174.24

## Physical Properties

Property code	Value	Unit	Source
gf	69.37	kJ/mol	Joback Method
hf	-148.48	kJ/mol	Joback Method
hfus	15.64	kJ/mol	Joback Method
hvap	50.24	kJ/mol	Joback Method
log10ws	-3.69		Crippen Method
logp	3.075		Crippen Method
mcvol	146.890	ml/mol	McGowan Method
pc	2832.35	kPa	Joback Method
rinpol	1509.00		NIST Webbook
ripol	2264.00		NIST Webbook
ripol	2264.00		NIST Webbook
ripol	2264.00		NIST Webbook
ripol	2266.00		NIST Webbook
ripol	2264.00		NIST Webbook
tb	589.43	K	Joback Method
tc	829.72	K	Joback Method
tf	359.10	K	Joback Method
vc	0.555	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	364.04	J/mol×K	589.43	Joback Method
cpg	381.43	J/mol×K	629.48	Joback Method
cpg	397.74	J/mol×K	669.53	Joback Method

cpg	413.00	J/mol×K	709.57	Joback Method
cpg	427.23	J/mol×K	749.62	Joback Method
cpg	440.46	J/mol×K	789.67	Joback Method
cpg	452.72	J/mol×K	829.72	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R228838&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R228838&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/75-422-0/4-7-Dimethyl-1-tetralone.pdf>

Generated by Cheméo on 2024-04-27 06:30:33.672846675 +0000 UTC m=+16488682.593423988.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.