

# Dibenzothiophene, 1,2,3,4-tetrahydro-

<b>Other names:</b>	1,2,3,4-Tetrahydrodibenzo[b,d]thiophene 1,2,3,4-tetrahydrodibenzothiophene Tetrahydrodibenzothiophene
<b>Inchi:</b>	InChI=1S/C12H12S/c1-3-7-11-9(5-1)10-6-2-4-8-12(10)13-11/h1,3,5,7H,2,4,6,8H2
<b>InchiKey:</b>	JCLPOPNXITXHOR-UHFFFAOYSA-N
<b>Formula:</b>	C12H12S
<b>SMILES:</b>	c1ccc2c3c(sc2c1)CCCC3
<b>Mol. weight [g/mol]:</b>	188.29
<b>CAS:</b>	16587-33-0

## Physical Properties

Property code	Value	Unit	Source
hfus	14.01	kJ/mol	Possible precursors and products of deep hydrodesulfurization of gasoline and distillate fuels III. The thermodynamic properties of 1,2,3,4-tetrahydrodibenzothiophene
hvap	75.30 ± 0.70	kJ/mol	NIST Webbook
log10ws	-4.53		Crippen Method
logp	3.780		Crippen Method
mcvol	146.510	ml/mol	McGowan Method
rinpol	294.30		NIST Webbook
rinpol	294.30		NIST Webbook

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hfust	32.03	kJ/mol	275.00	NIST Webbook
hvapt	67.30 ± 0.30	kJ/mol	480.00	NIST Webbook
hvapt	64.50 ± 0.20	kJ/mol	480.00	NIST Webbook
hvapt	61.80 ± 0.20	kJ/mol	480.00	NIST Webbook
hvapt	59.20 ± 0.30	kJ/mol	480.00	NIST Webbook
hvapt	56.50 ± 0.40	kJ/mol	480.00	NIST Webbook
hvapt	70.30 ± 0.30	kJ/mol	480.00	NIST Webbook

# Sources

Possible precursors and products of deep hydrodesulfurization of gasoline	<a href="https://www.doi.org/10.1016/j.jct.2003.12.012">https://www.doi.org/10.1016/j.jct.2003.12.012</a>
McGowan Method III. The thermodynamic properties of 1,2,3,4-tetrahydrodibenzothiophene:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
Crippen Method:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C16587330&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C16587330&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

# Legend

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>rinpol:</b>	Non-polar retention indices

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